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Qin, N., Li, L., Wan, X. et al (2024). Increased CO₂ fixation enables high carbon-yield production of 3-hydroxypropionic acid in yeast. *Nature Communications*, 15(1). <http://dx.doi.org/10.1038/s41467-024-45557-9>

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Increased CO₂ fixation enables high carbon-yield production of 3-hydroxypropionic acid in yeast

Received: 31 August 2023

Accepted: 28 January 2024

Published online: 21 February 2024

Check for updates

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CO₂ fixation plays a key role to make biobased production cost competitive. Here, we use 3-hydroxypropionic acid (3-HP) to showcase how CO₂ fixation enables approaching theoretical-yield production. Using genome-scale metabolic models to calculate the production envelope, we demonstrate that the provision of bicarbonate, formed from CO₂, restricts previous attempts for high yield production of 3-HP. We thus develop multiple strategies for bicarbonate uptake, including the identification of Sul1 as a potential bicarbonate transporter, domain swapping of malonyl-CoA reductase, identification of Esbp6 as a potential 3-HP exporter, and deletion of Uga1 to prevent 3-HP degradation. The combined rational engineering increases 3-HP production from 0.14 g/L to 11.25 g/L in shake flask using 20 g/L glucose, approaching the maximum theoretical yield with concurrent biomass formation. The engineered yeast forms the basis for commercialization of bio-acrylic acid, while our CO₂ fixation strategies pave the way for CO₂ being used as the sole carbon source.

Biobased chemical¹ and biofuel² production as replacement of petroleum-based production could assist in mitigating green-house gas emissions while establishing sustainable economic activities³. However, these biobased productions also result in emissions of the greenhouse gas CO₂, and fixing emitted CO₂ into products is therefore of great interest⁴, in particular as this may also result in higher carbon-yields and therefore improves process economics.

Intracellular inorganic carbon exists in four forms: dissolved CO₂, carbonic acid, bicarbonate, and carbonate⁵, that could be interconverted based on the pH and salinity. It has been reported

that compared with 5% CO₂ supply, 10% CO₂ could increase itaconic acid production by two-fold in yeast⁶. At pH 7.4 and 20 °C, the concentration of CO₂ in equilibrium with the gaseous state is only 0.012 mM, whereas the concentration of bicarbonate in equilibrium with air is 0.26 mM⁷. Since the pH in the cytoplasm is approximately neutral⁸, inorganic carbon mainly exists in the form of bicarbonate⁹, which means increasing intracellular bicarbonate concentrations could be more appealing than enhancing the higher CO₂ partial pressure. Bicarbonate can be captured by native carboxylation reactions, of which the acetyl-CoA carboxylation catalysed by acetyl-

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CoA carboxylase Acc1 is known to carry a high flux in yeast. The generated malonyl-CoA can be transformed into 3-hydroxypropionic acid (3-HP), which represents a valuable chemical intermediate. Efficient capture of CO₂ into 3-HP can improve the carbon yield in its production, and this step is also crucial in the native CO₂ fixation pathway 3-hydroxypropionate/4-hydroxybutyrate (HP/HB) cycle¹⁰ and 3-hydroxypropionate bi-cycle¹¹.

3-HP holds a significant position as the potentially third-largest biobased platform compound, with an estimated potential annual market size of USD 10 billion^{12,13}. One of the key reasons behind its prominence is its ease of dehydration into acrylate, a crucial component used in the production of synthetic polymers like super-absorbents applied in disposable diapers¹⁴. The production of 3-HP has thus garnered considerable attention, but the major hurdle remains the low yield of 3-HP on the substrate, posing a significant challenge to establish an industrial scale process¹⁵. A comprehensive analysis of 3-HP production in yeast reveals that the best yield achieved thus far in terms of the 3-HP to substrate (glucose) ratio is approximately 0.31 g/g¹⁶, which corresponds to only 39.7% of the maximum theoretical yield¹⁵. This falls short of the commercial requirement for 3-HP to acrylate production, which highlights the current limitations in 3-HP production and underscores the need for significant improvements in yield to meet the demands of the industry.

In this study, we conduct production envelope analysis by Yeast8 model and find that previous attempts to produce 3-HP is restricted by the availability of bicarbonate. We therefore engineer yeast to break this restriction and hereby achieve high carbon yield production of 3-HP. This is achieved through employing multiple strategies to increase the accessible bicarbonate, minimize native CO₂ release, as well as by avoiding carbon waste. Hereby we achieve a 0.5625 g 3-HP/g glucose yield (89.3% of the theoretical yield of 3-HP with biomass generation). Furthermore, the high titre of 3-HP obtained through a microbial fermentation can easily be transformed to acrylate with a yield of 100%.

Results

Limited provision of bicarbonate restricts 3-HP production

The production envelope analysis of 3-HP via the malonyl-CoA reductase pathway was calculated using the Yeast8 model¹⁷, and this was used to analyse the carbon flux and identify limiting factors for 3-HP production. Contrary to the general belief that the precursors of acetyl-CoA and malonyl-CoA, as well as the redox cofactors and ATP are the key requirements for 3-HP production, our analysis suggested that a high flux of bicarbonate (Fig. 1a) could be crucial for efficient 3-HP production, i.e., the flux of bicarbonate should be 1.562 times higher than that of the substrate (glucose) to achieve the maximum 3-HP production (Fig. 1a). Unfortunately, such a flux of bicarbonate is unattainable within the native metabolism, even if we assume that bicarbonate can be freely converted from CO₂. The cellular metabolism is simply incapable of providing enough CO₂ to generate the required bicarbonate for 3-HP production¹⁸, regardless of the metabolic state being respiration (efficient CO₂ generation state) or fermentation (non-efficient CO₂ generation state) (Fig. 1a). Moreover, if we consider that the CO₂ generated is in a gaseous state intracellularly, it would promptly leave the liquid phase, and the carboxylation process by Acc1 necessitates the presence of bicarbonate dissolved in water (Fig. 1b). These factors further exacerbate the deficiency of bicarbonate. Specifically, in the native metabolic context, the limited flux of bicarbonate significantly reduced the theoretical maximum yield of 3-HP from 0.781 g/g glucose to approximately 0.440 g/g (fermentation state)–0.490 g/g (respiration state) (Fig. 1a). This result indicated that the availability of bicarbonate is the primary bottleneck for 3-HP production, effectively restricted all previous attempts at metabolic flux rewiring in this production envelope boundary. To address this issue, it is imperative to ensure an ample supply of

bicarbonate for Acc1 carboxylation while minimizing CO₂ loss during acetyl-CoA and energy generation.

The carbon conserved loop from glucose to 3-HP

We started by establishing a carbon conserved loop that effectively recycled the lost carbon to increase the theoretical carbon-yield for 3-HP production (Fig. 1b). The split mutant of the bi-functional enzyme MCR from *Chloroflexus aurantiacus*¹⁹ was used for 3-HP production, as it exhibited efficient conversion of malonyl-CoA to malonic semi-aldehyde (MSA) via the mutated MCR-C^{N941V K1107W S115R} domain and then to 3-HP via the MCR-N domain²⁰ (Fig. 1b). Integration of the split MCR into the chromosome together with *ACC1* overexpression resulted in a 3-HP titre of 0.136 g/L in the QLW1 strain (Fig. 1c). Then, to test our hypothesis we added 10 mM sodium bicarbonate to the medium, and indeed increased the 3-HP titre to 0.15 g/L (*P* value = 0.012813). Thus, for the following fermentation processes, 10 mM sodium bicarbonate was consistently added unless described otherwise.

Next, to reduce the CO₂ release while producing acetyl-CoA, we integrated the phosphoketolase (PK) pathway reported in our previous study²¹, including a xylulose-5-phosphate-preferred phosphoketolase (xPK) from *Leuconostoc mesenteroides* and a phosphotransacetylase (PTA), from *Clostridium kluyveri*, and enhanced 3-HP production to 0.39 g/L (Fig. 1c, QLW2). Further, to release Acc1 from post-translational regulation, we introduced two site mutations, S659A and S1157A, which prevent phosphorylation of Acc1 by Snf1, thereby activating Acc1's activity²². It then resulted in an increased 3-HP titre to 0.65 g/L in the QLW3 strain (Fig. 1c).

To achieve a higher flux through the PK pathway, extensive genotype modifications were reported to redirect carbon flux from glycolysis to the oxidative pentose phosphate (oxPP) pathway²¹. We instead turned to transcription factors (TFs) to investigate whether a single gene modification could achieve the overall carbon flux rewiring. Stb5 is a TF and an NADPH regulator²³ that represses the expression of *PGII*, coding the rate-limiting reaction of glycolysis at glucose-6-phosphate to rewire the flux from glycolysis to the oxPP pathway²⁴. Stb5 has been used to redirect the carbon flux for the production of protopanaxadiol in yeast²⁵. We thus overexpressed *STB5* with the *TEF1* promoter. However, the growth of the strain was significantly inhibited (Supplementary Fig. 1). qPCR results indicated that the strength of the *TEF1* promoter was around 12-fold higher than that of the native *STB5* promoter (Fig. 1d, QLW4), that may have been too strong for regulating the TF expression in the cell. Thus, we placed the TATA box adjacent to the core promoter of *TEF1*²⁶ (Fig. 1d) and constructed an artificial promoter to replace the native promoter of *STB5*, and qPCR results showed that its strength was around 2.5-fold greater than that of the native *STB5* promoter (Fig. 1d, QLW5). Indeed, the strain with this artificial promoter-controlled *STB5* enhanced 3-HP production to 0.74 g/L (Fig. 1c, QLW5).

Meanwhile, we tried to employ carbonic anhydrase and a bicarbonate transporter to enhance the intracellular bicarbonate content for the production of 3-HP (Fig. 1e). This CO₂ concentrating system has successfully increased the production of succinate in *E. coli*^{27,28}. Similarly, overexpression of *NCE103*, the native carbonic anhydrase in yeast²⁹, could also improve 3-HP production to 0.7 g/L in QLW6 (Fig. 1f). Furthermore, we tried to find a native bicarbonate transporter in yeast. Bicarbonate diffuses poorly into the cell²⁷, and therefore requires SLC4 and SLC26 as two kinds of anion transporters to assist the active transport process³⁰. Bor1, a boron efflux transporter of the plasma membrane belonging to the SLC4 family, can also bind bicarbonate (Fig. 1g)³¹, whereas no bicarbonate transporter from the SLC26 family has been reported in yeast³². To identify potential bicarbonate transporters, we performed a phylogenetic analysis of key species of the SLC4 and SLC26 families (Fig. 1g). We identified Sul1/2, which were previously regarded as sulphate permeases³³. Bor1 is an efflux transporter, while Sul1/2 are influx transporters. Yeast Sul1 and Sul2 share a

Fig. 1 | Production envelope analysis and bicarbonate metabolism rewiring for CO₂ fixation. **a** The production envelope analysis for 3-HP production. Different colour represented the range of 3-HP production with its concurrent biomass under different bicarbonate flux. Reports used the malonyl-CoA pathway^{16,21,22,102–106}, the oxaloacetate pathway⁴⁴, and the β -alanine pathway⁸² were labelled with dot, square, and diamond. The q_s indicated the consumed rate of glucose. The DCW was either measured directly or estimated based on OD₆₀₀ (DCW equals to 70% of OD₆₀₀). **b** The malonyl-CoA reductase pathway was used to produce 3-HP. Transcription factor Stb5 could realize the carbon flux rewiring from glycolysis to the oxPP pathway. **c** Adjusting *STB5* expression by *ARTp* could together with the expression optimization of MCR domains, Acc1 and the PK pathway enhance 3-HP production to 0.74 g/L. **d** Substitution of the native promoter of *STB5* with the *TEF1p* and an artificial promoter *ARTp*, respectively, and quantification of the *STB5* with *TEF1p* and *ARTp* strength by qPCR. **e** Cellular bicarbonate derived from the potential bicarbonate transporter Sul1 (into the cell), Bor1 (outside the

cell), and the endogenous synthesis from CO₂ by native carbonic anhydrase Nce103. **f** Providing more bicarbonate by overexpressing *SUL1* enhanced 3-HP production to 0.83 g/L. **g** Phylogenetic analysis of SLC4 and SLC26 families identified Bor1, Sul1, and Sul2 as potential bicarbonate transporters in *S. cerevisiae*. **(h)** Rewiring the carbon flux from OAA to the high production of 3-HP. Overexpressing the phosphatase Ptc7 could significantly increase the 3-HP production based on introducing Cit1^{S462A} mutation, overexpressing *CIT1*, *YHM2*, and *IDP2*. **(i)** Distribution of CO₂ releasing reactions and bicarbonate utilizing reactions in *S. cerevisiae*. Abbreviations were defined in Supplementary Data 5 and 6. 10 mM sodium bicarbonate was added to the defined minimal medium was used for the 3-HP production. All data were presented as mean \pm SD of biological triplicates. Significant comparisons of two groups were indicated in the graphs statistical analysis performed using a two-tailed unpaired Student's *t*-test (*p* < 0.05, *******p* < 0.01, ********p* < 0.001). Source data are provided as a Source Data file.

the importance of intracellular bicarbonate availability for 3-HP production.

Next, we intended to combine the above strategies of *STB5* and *SUL1* overexpression to close this carbon conservation loop, yet the 3-HP production in the resulted QLW10 strain dropped to 0.31 g/L, which was even lower than that of the reference strains QLW5 and QLW8 (Fig. 1h). Analysis of extracellular metabolites revealed the presence of 0.37 g/L of oxaloacetate (OAA) in the QLW10 strain but not the QLW8 strain (Supplementary Fig. 4). We speculated that the increased intracellular bicarbonate content also improved other bicarbonate utilizing reactions, such as the carboxylation of pyruvate to OAA catalysed by Pyc1 and Pyc2, and therefore tried various strategies to decrease the accumulation of OAA (Fig. 1i). Yet, the deletion of *CIT2* that encoded the key gene of the glyoxylate cycle didn't recover 3-HP titre (Fig. 1h, QLW11), with acetate accumulated instead (Supplementary Fig. 5). We next focused on enhancing the utilization of OAA. Firstly, we overexpressed *CIT1* encoding the mitochondrial citrate synthase that converts OAA and acetyl-CoA to citrate, and improved 3-HP production to 0.44 g/L (Fig. 1h, QLW12). Similarly, we tested Cit1^{S462A} enabling the enzyme to defer phosphorylation regulation³⁵, and increased 3-HP production to 0.45 g/L (Fig. 1h, QLW13). We also overexpressed Ptc6 and Ptc7, which could selectively dephosphorylate mitochondrial proteins, especially the Ptc7 that could dephosphorylate Cit1 and enhance yeast respiration³⁶. Indeed, overexpression of *PTC7* in the QLW15 strain could not only recover but further increase the production of 3-HP to 1.18 g/L (Fig. 1h). Next, we overexpressed Yhm2, an OAA transporter in the mitochondrial membrane³⁷, and the cytosolic NADP-specific isocitrate dehydrogenase Idp2 that could provide more cytosolic NADPH³⁷, further increased 3-HP production to 1.34 g/L (Fig. 1h, QLW19), that were 81.1% and 60.9% improvement over the control strain QLW5 and QLW8, respectively.

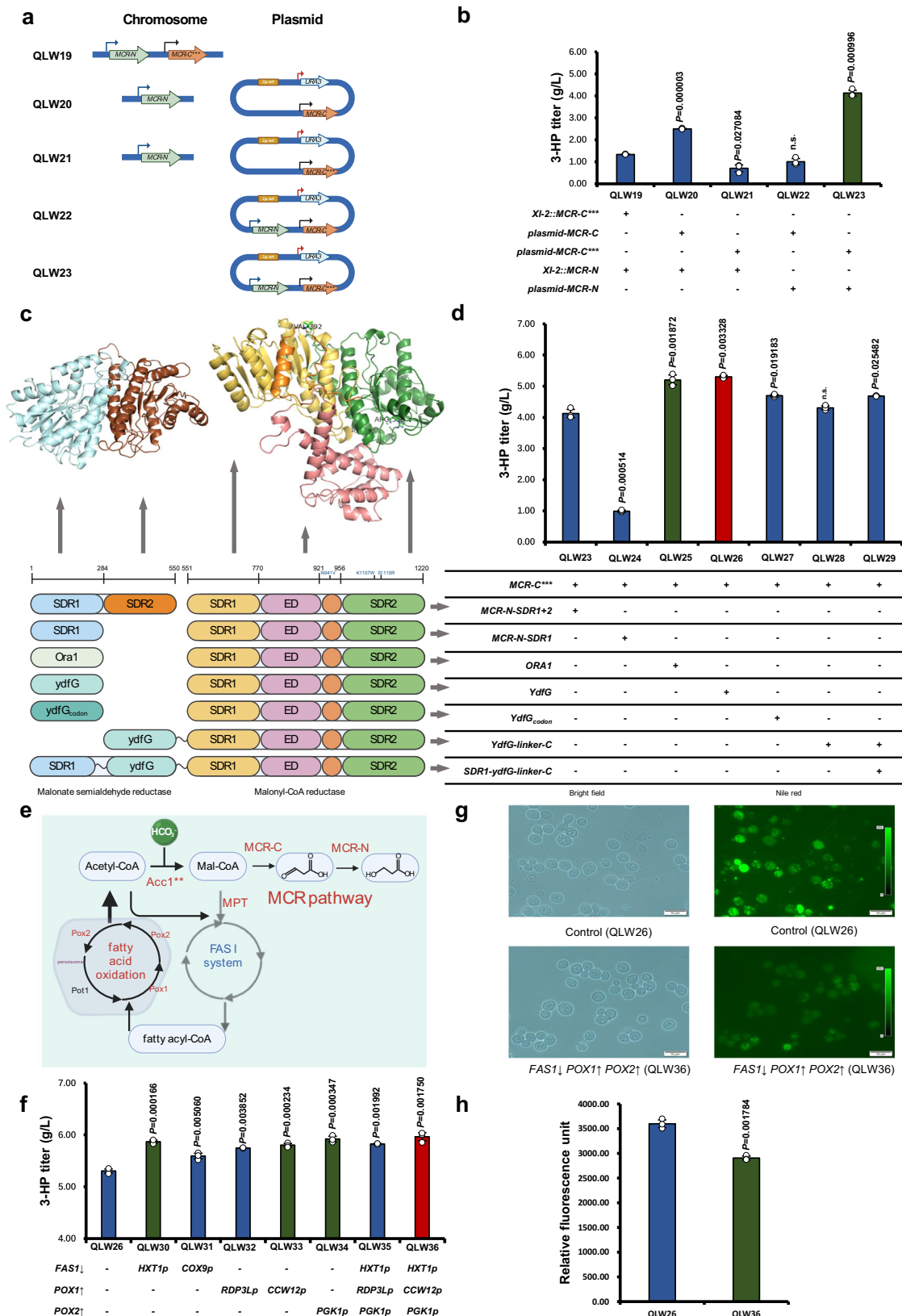
Improved conversion of malonyl-CoA to 3-HP

To provide enough malonyl-CoA for 3-HP production, another copy of *ACCI*^{S659A} *S1157A* with *TEF1* promoter was integrated into the chromosome, yet the growth of the strain was strongly disturbed. It has been reported that the intracellular malonyl-CoA level is strictly regulated by the AMP-activated kinase (AMPK)³⁸ signalling pathway and a high concentration of malonyl-CoA could be toxic to the cell³⁹. This result suggested that the elevated malonyl-CoA concentration being beyond the range the cell could handle⁴⁰. Therefore, we focused on more efficient conversion of malonyl-CoA into 3-HP.

We started by expressing MCR-N and MCR-C in different combinations, and expressing the MCR-N domain and MCR-C^{N941V} *K1107W* *S1115R* domain^{16,20} using a higher copy number plasmid increased 3-HP production to 4.12 g/L in QLW23 (Fig. 2a, b). Currently, there is no crystal structure of MCR from *C. aurantiacus*, and only crystal structures of the MCR-N domain and the MCR-C domain from *Porphyrobacter dokdonensis* have been published⁴¹. Using SWISS-MODEL, we generated a homologous model of MCR from *C. aurantiacus* using known structures

of the MCR-N domain (PDB: 6k8v) and MCR-C domain (PDB: 6k8s) from *P. dokdonensis* (Fig. 2c) as the template. We noticed that the MCR-N domain could be separated into two short-chain dehydrogenase/reductase⁴¹ domains: SDR domain 1 (MCR-N domain/SDR1, M1–L284) and SDR domain 2 (MCR-N domain/SDR2, P285–I550). The MCR structure revealed that the cofactor and substrate only bind to the SDR1 domain, indicating that enzyme catalysis may not occur at the SDR2 of the MCR-N domain⁴¹. Yet, when we deleted the MCR-N-SDR2 domain, the production of 3-HP had a strong decrease (Fig. 2d, QLW24). This result suggested that even though the MCR-N-SDR2 domain has no catalytic function, it may help the MCR-N-SDR1 domain to maintain its proper structure. To find a more efficient domain converting MSA to 3-HP, we performed domain swapping using SDRs from different species and obtained the improved malonyl-CoA reductase. Specially, we identified potential MCR-N-SDR1 domains through the substrate structural similarity, such as Ora1 (reported as the serine dehydrogenase)⁴² from *S. cerevisiae*, and YdfG (reported as the NADP⁺-dependent dehydrogenase with broad substrate specificity on 3-hydroxy acids)⁴² from *E. coli* (Fig. 2c). Serine can be seen as 2-amino-3-hydroxypropionic acid, which could be regarded as the amino group substituted at the second carbon atom site of 3-HP. The substrate structural similarity together with the broad substrate specificity to 3-hydroxy acids of Ora1 and YdfG indicated the potential of these enzymes for 3-HP production. Indeed, the use of Ora1 and YdfG as MCR-N-SDR1 exhibited 26.1% and 28.5% improvements, producing 5.2 g/L and 5.3 g/L 3-HP, respectively (Fig. 2d, QLW25, QLW26). Neither codon optimization nor the application of a protein linker⁴³ could further improve the production of 3-HP (Fig. 2d). The finding that common cell factories such as *S. cerevisiae* and *E. coli* possess native enzymes catalysing malonic semialdehyde to 3-HP could potentially explain why adjusting the expression the of MCR-N domain and MCR-C domain was always irregular and been impute into ambiguity explanations^{20,44}.

Meanwhile, since malonyl-CoA was also utilized for fatty acid synthesis in the cell⁴⁵, we tried to reduce the generation of fatty acids to increase the production of 3-HP (Fig. 2e). Specially, we used the *COX9* promoter, which is a weaker promoter than the native *FASI* promoter, and *HXT1* promoter, which is repressed by glucose, to control the expression of *FASI*, respectively⁴⁶ (Fig. 2f). Using this approach, we were able to increase the titre of 3-HP to 5.87 g/L (Fig. 2f, QLW30). Meanwhile, we also up-regulated the β -oxidation pathway to decompose fatty acids back to NADPH and acetyl-CoA for 3-HP production (Fig. 2e). The resulting the QLW33 strain overexpressing *POX1* that encodes fatty-acyl coenzyme A oxidase⁴⁷ improved 3-HP production to 5.8 g/L. Similarly, the QLW34 strain overexpressing *POX2* that encodes 3-hydroxyacyl-CoA dehydrogenase and enoyl-CoA hydratase⁴⁷, improved 3-HP production to 5.92 g/L (Fig. 2f). We thus combined these strategies, yet did not observe a superimposed effect, with the 3-HP production in QLW36 fluctuated around 6 g/L (Fig. 2f). To verify the repression of lipid synthesis, we used Nile red that



specifically binds to neutral lipids to characterize the yeast. The results showed a significant reduction in neutral lipids in the QLW36 strain compared with that of the control strain QLW26 (Fig. 2g, h), indicated a successful rewiring of the carbon flux away from fatty acids. That the titre of 3-HP could not be significantly increased may be because of other bottlenecks in the metabolism.

Increasing 3-HP tolerance and the identification of a permease for transport of 3-HP

The 3-HP transport and toxicity represent a bottleneck for further improving 3-HP production in the QLW36 strain. Our experiments showed that after 72 h of fermentation with 5.5–6 g/L of 3-HP generated (strains QLW30 to QLW36), the pH of the medium dropped

Fig. 2 | Enzyme and pathway engineering for efficient production of 3-HP.

a Genes encoding of split MCR enzymes were either integrated in the chromosome or expressed using the plasmid system. **b** MCR-N domain and MCR-C*** domain co-expressed with the high copy plasmid could produce 4.12 g/L 3-HP. **c** Homologous modelling structure of the domains in malonate semialdehyde reductase and malonyl-CoA reductase from *C. aurantiacus*. **d** Swapping of native MCR domains with Ora1 and YdfG improved 3-HP production to 5.2 g/L and 5.3 g/L, respectively. **e** Carbon flux rewiring from fatty acids to 3-HP. **f** Downregulation of *FAS1* combined with upregulation of *POX2* and *POX1* improved the production of 3-HP. **g** Nile red staining demonstrated that the size and the number of lipid droplets decreased in the fatty acid oxidation strain QLW36 compared with that of QLW26. **(h)** The

fluorescence intensity of neutral lipids stained with Nile red in QLW26 and QLW36 was quantified in Relative Fluorescence Units (RFU). The RFU values were corrected by subtracting both the inherent autofluorescence of the samples and the fluorescence contributed by the solvent in the presence of Nile red (blank). Abbreviations were defined in Supplementary Data 5 and 6. The defined minimal medium with 10 mM NaHCO₃ was used for the 3-HP production. All data were presented as mean ± SD of biological triplicates. Significant comparisons of two groups were indicated in the graphs statistical analysis performed using a two-tailed unpaired Student's *t*-test (**p* < 0.05, ***p* < 0.01, ****p* < 0.001). Source data are provided as a Source Data file.

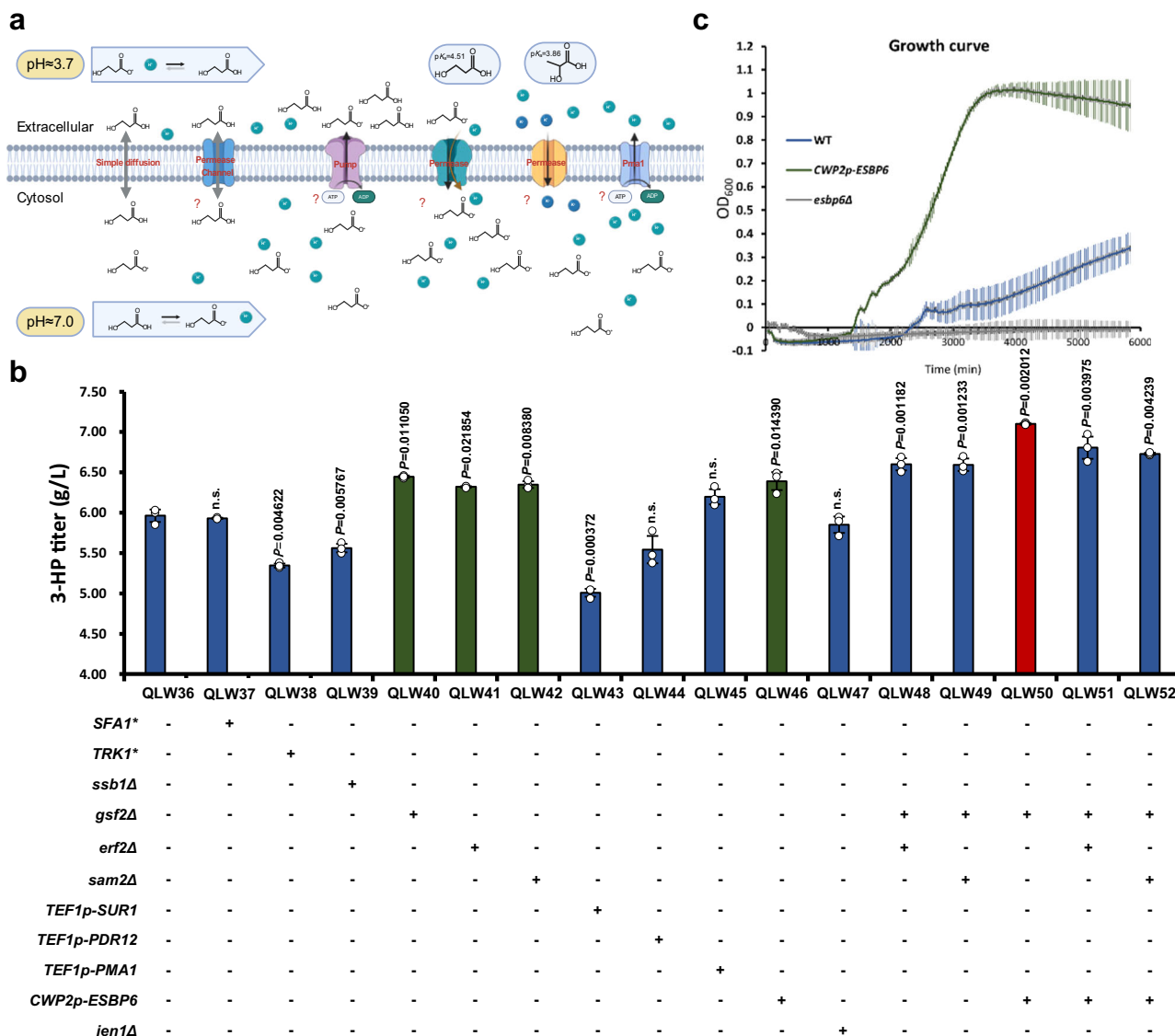


Fig. 3 | Genotype identification of 3-HP tolerance and increase in the production of 3-HP. **a** Schematic representation of transporters and permeases in the cell membrane. **b** Overexpression of *ESBP6*, deletion of *SAM2*, *GSF2*, or *ERF2* could significantly improve the production of 3-HP. **c** Growth profiling showed that overexpression of *ESBP6* could remarkably increase cell tolerance to high concentrations of 3-HP, compared with the WT strain. Abbreviations were defined in

Supplementary Data 5 and 6. The defined minimal medium with 10 mM NaHCO₃ was used for the 3-HP production. All data were presented as mean ± SD of biological triplicates. Significant comparisons of two groups were indicated in the graphs statistical analysis performed using a two-tailed unpaired Student's *t*-test (**p* < 0.05, ***p* < 0.01, ****p* < 0.001). Source data are provided as a Source Data file.

below 4. The pK_a value of 3-HP is 4.51, and under this pH value, 3-HP remains predominantly undissociated in the medium (Fig. 3a). Free undissociated 3-HP can easily penetrate the cell membrane through diffusion and then release protons to become anions in the cytoplasm where the pH is around neutral⁸. The charged nature of the

anion cannot diffuse back out of the cell, resulting in its accumulation in the cytosol that could cause intracellular acidification, which is toxic for the cell⁴⁸. The proton and anion must be pumped out of the cell, and this process is ATP-dependent through the H⁺-ATPase and ATP-binding cassette (ABC) pumps⁴⁸. This process consumes a

large amount of ATP and can result in ATP depletion, leading to a decrease in the overall 3-HP yield.

We started by testing mutations to improve yeast tolerance of 3-HP (Fig. 3b). For example, the Sfa1 bifunctional alcohol dehydrogenase and the Sfa1^{C276S} mutant have been reported to improve 3-HP tolerance in *S. cerevisiae*⁴⁹. Trk1, a component of the Trk1-Trk2 potassium transport system, was reported to improve yeast tolerance to propionate by increasing potassium influx through the Trk1^{Al169E} mutation⁵⁰. Yet, these two mutations could not increase the production of 3-HP (Fig. 3b, QLW37 and QLW38). Moreover, lactate (pKa 3.86) and 3-HP (pKa 4.51) are structural isomers that have been found to exhibit the same degree of growth inhibition at the same undissociated free acid concentration^{51,52}, indicating that the mutation involved in the tolerance of lactate may have the potential to improve 3-HP tolerance. The deletion of the ribosome-associated chaperone Ssb1 was reported to improve yeast tolerance to lactate⁵³. The deletion of *GSF2* could increase lactate production⁵⁴ by releasing the Crabtree effect of yeast⁵⁵, leading to upregulation of gene expression involved in respiration, which has a higher ATP yield than fermentation and is crucial to organic acid production⁵⁶. Deletion of *ERF2*, a subunit of a palmitoyl transferase, could increase lactate production but not lactate tolerance in yeast⁵⁷. Deletion of *SAM2*, which catalyses the synthesis of the cellular cofactor S-adenosylmethionine involved in phospholipid biosynthesis and membrane remodelling during acid stress, was also reported to increase lactate production in yeast⁵⁸. Furthermore, the Sur1^{I245S} mutation and the overexpression could increase lactate tolerance and production⁵⁷. Indeed, we demonstrated that the deletion of either *GSF2* (QLW40), *ERF2* (QLW41), or *SAM2* (QLW42) significantly improved the production of 3-HP, to the levels of 6.45, 6.32, and 6.35 g/L, respectively (Fig. 3b).

So far, no 3-HP exporter has been reported. Based on the fact that lactate and 3-HP are structural isomers, we next turned to the identification of potential 3-HP exporters based on previous lactate reports. Plasma membrane ABC transporter Pdr12, which could transport carboxylic acids with an aliphatic chain ranging from one to seven⁵⁹ at the plasma membrane. P2-type H⁺-ATPase Pma1⁶⁰, which controls the intracellular pH, have been reported to improve lactate tolerance in yeast⁶¹. The overexpression of *ESBP6*, a protein with similarity to monocarboxylate permeases, was reported to improved lactate resistance and production in *S. cerevisiae*⁶². Meanwhile, the deletion of Jen1, the monocarboxylate/proton symporter in the plasma membrane⁶³, could increase the production of lactate in yeast^{54,64,65}. Indeed, compared with that of the control strain QLW36, the overexpression of *ESBP6* improved the production of 3-HP to 6.39 g/L (Fig. 3b, QLW46). To verify the potential mechanism by which *Esbp6* improved 3-HP production, *ESBP6* was overexpressed in the wild type (WT) strain, and could significantly increase cell tolerance compared with WT in a medium containing 50 g/L 3-HP at pH 3.5 (buffered with potassium citrate), while the deletion of *ESBP6* almost completely abolished growth in this medium (Fig. 3c). Moreover, the overexpression of *ESBP6* in the WT strain led to a notable reduction in intracellular 3-HP content compared with that of the WT strain (Supplementary Fig. 6). These results indicated that *Esbp6* could help exporting 3-HP out of the cell, thereby increasing the tolerance of yeast and improving 3-HP production. To our knowledge, this could be the evidence for potential 3-HP exporters. Notably, combining the overexpression of *ESBP6* with the deletion of *GSF2* improved 3-HP production to 7.1 g/L in strain QLW50 (Fig. 3b).

Conserved bicarbonate pool for 3-HP production

There are mainly three enzymes utilizing bicarbonate in yeast (Fig. 1i): Acc1 fixes bicarbonate with acetyl-CoA to form malonyl-CoA, Pyc1/2 fixes bicarbonate to transform pyruvate into OAA, as well as Cpa1/2 fixes bicarbonate to generate carbamoyl-phosphate⁶⁶, and then to the pyrimidine ring and arginine (Fig. 4a). It has been demonstrated that

bicarbonate limitation in the cytoplasm could restricted de novo nucleotide synthesis, which inhibits growth of cancer cells (the Weinberg effect)⁶⁷. This report enlightens that the potential competition for bicarbonate between the de novo synthesis of the pyrimidine ring and malonyl-CoA may reduce 3-HP production (Fig. 4a). The plasmid used to express *MCR-C* and *YdfG* had a high copy number with a *URA3* marker. This means that the strain with this plasmid will have a stronger expression of *URA3*, a key step in pyrimidine ring synthesis, leading to the potential competition with Acc1 for bicarbonate. To conserve bicarbonate to 3-HP production, we replaced the *URA3* marker with the *HIS3* marker that is not involve in the bicarbonate metabolism⁶⁸ (Fig. 4a). Indeed, the *HIS3* marker plasmid significantly enhanced 3-HP production from 7.1 g/L to 8.22 g/L (Fig. 4b, QLW53), while the qPCR result even exhibited a moderate decrease in the plasmid copy number upon the change in the marker gene (Supplementary Fig. 7). To demonstrate the improvement of 3-HP production by auxotrophic marker swapping was caused by the decrease in the competition of bicarbonate for the de novo uracil synthesis, we added isotope labelled NaH¹³CO₃ to the medium and tested the M+1 presence of uracil in strains with different auxotrophic-marker plasmids. Indeed, consistent with the significantly increased 3-HP production, the M+1 presence of uracil was significantly decreased in the QLW53 strain compared with that of the QLW50 strain (Fig. 4b, c). These results confirmed that the auxotrophic marker swapping increased the 3-HP production by releasing the 3-HP generation from the bicarbonate competition with the de novo uracil synthesis (Fig. 4c). Different auxotrophs have been shown to impact yeast metabolism⁶⁹, which can affect the production of fatty acids⁷⁰. Yet, mechanisms behind this impact have not been fully understood⁷⁰. Here, we demonstrated that the distinct bicarbonate metabolic backgrounds of these auxotrophic marker turbulent metabolic capacities of yeast⁷¹, thereby increasing the production of 3-HP.

The results of the enhanced 3-HP production with the *HIS3* marker plasmid suggested that cytoplasmic bicarbonate generated from dissolved CO₂ could be limited due to its low solubility. To provide enough bicarbonate for malonyl-CoA generation, we supplemented the medium with 75 mM (0.15 g/flask) calcium carbonate, which further enhanced 3-HP production from 8.22 to 9.12 g/L (Fig. 4d and Supplementary Fig. 8). To test whether the increase in 3-HP production was due to the addition of bicarbonate rather than pH buffering, we conducted a production of 3-HP experiment with 2-(*N*-morpholino) ethane sulfonic acid (MES) that could buffer the pH to the same level as calcium carbonate did. The results showed that the titre of the fermentation with MES buffering was only 8.16 g/L, which was not significantly different from the control (Fig. 4d). We have also tested different concentrations of calcium hydroxide, and only 40 mM (0.0592 g/flask) calcium hydroxide could slightly improve 3-HP production to 8.41 g/L (Supplementary Fig. 9), whereas when further increased calcium hydroxide concentrations the cells stopped growth. A similar strategy has been utilized to increase succinate production in industry, yet the mechanism underneath was not indicated^{72,73}. Here by model calculation and experiment demonstration, we suggested that it was because the limited intracellular CO₂ concentration as well as the thermodynamic and kinetic favouring of the bicarbonate fixation reaction that increased the production.

Inositol phosphate application provides more ATP for 3-HP production

The positive effect on 3-HP production by deleting *GSF2* indicated that the energy was crucial for the 3-HP yield. If we consider the energy demand for 3-HP transmembrane during production, this ATP deficiency would be more intense. We thus selected different strategies to provide more ATP for 3-HP production. *S. cerevisiae* has the Crabtree effect, which produces an excess amount of ethanol in high

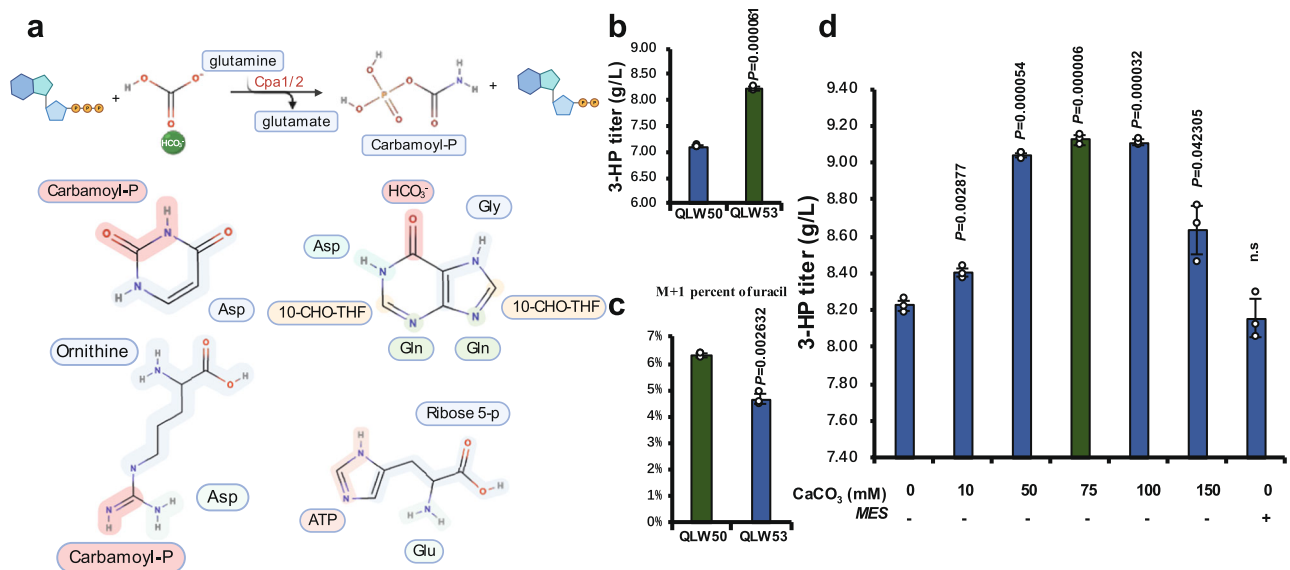


Fig. 4 | Alleviating bicarbonate competitions from the de novo pyridine ring synthesis enhanced 3-HP production. **a** Carbamoyl-P was synthesized by Cpa1/2 from ATP, bicarbonate, and glutamate. Atom sources for purine, pyrimidine, arginine, and histidine were marked. **b** Substitution of *URA3* marker by *HIS3* marker in the same plasmid background increased the 3-HP titre to 8.22 g/L in the QLW53 strain. **c** ¹³C isotope labelling indicated that less M + 1 uracil was produced in QLW53 strain corresponding to the higher 3-HP titre. **d** CaCO₃ was used to provide more bicarbonate recycling CO₂, 0, 10, 50, 75, 100, and 150 mM represent 0 g/flask,

0.02 g/flask, 0.10 g/flask, 0.15 g/flask, 0.20 g/flask, and 0.30 g/flask. Abbreviations were defined in Supplementary Data 5 and 6. The defined minimal medium with 30 mM ¹³C isotope labelling NaHCO₃ or the corresponding concentrations of CaCO₃ as shown in the figures was used for the 3-HP production. All data were presented as mean ± SD of biological triplicates. Significant comparisons of two groups were indicated in the graphs statistical analysis performed using a two-tailed unpaired Student's *t*-test (**p* < 0.05, ***p* < 0.01, ****p* < 0.001). Source data are provided as a Source Data file.

concentrations of glucose and reduces energy efficiency^{74,75}. Moreover, the P/O ratio of *S. cerevisiae* is around 1, which indicates that one mole of glucose could only generate approximately 16 moles of ATP (oxidation phosphorylation plus substrate phosphorylation)⁷⁶. Thus, rebalancing central metabolism and enhancing the ATP yield from glucose could be important.

After glucose is transported inside the cell, it would be phosphorylated by Hxk2 or Glk1 and Hxk1, which restricted its transportation out of the cell⁷⁷ (Fig. 5a). Deleting main functional carrier *HXX2* can reduce the glucose consumption rate, which could release the Crabtree effect to get a higher ATP yield⁷⁸. Similarly, Mth1 is involved in the glucose signalling pathway, and its mutant Mth1^{Δ231} acts as a transcription factor to repress hexose transporters (Fig. 5a), that ultimately restricts the Crabtree effect and derepress respiration⁷⁸. Meanwhile, the mutation of Med2^{432Y}, which is the tail module of RNA polymerase II mediator, can decrease the expression of genes involved in glycolysis and increase the expression of genes involved in protein synthesis. These mutations in principle may be able to release more ATP⁷⁹, yet did not increase the production of 3-HP in strain QLW53 any further (Fig. 5b). The glucose repression on the respiratory pathway may still be too strong that glycolysis and respiration may not be well balanced for high 3-HP production. We next turned to our recently identified Oca5 that degrade 5-diphosphoinositol 1,2,3,4,6-pentakisphosphate (5-InsP₇)⁷⁸, which serves as an energy sensor to sense ATP concentrations and balances gene expressions in glycolysis and respiration. Indeed, *OCA5* deletion significantly increased 3-HP production from 9.12 g/L in QLW53 to 10.15 g/L in QLW58 (Fig. 5b). To verify that the production enhancement was because of the rebalance of glycolysis and respiration rather than simply enhanced respiration, we tested the cultivation in baffled shake flask that could provide more oxygen for oxidative phosphorylation, and indeed the 3-HP production did not improve for QLW53 (Fig. 5b). We also tested the deletion of *OCA3*⁸⁰ and upregulation of *KCSI*⁸¹, respectively, that could also increase 5-InsP₇ concentrations, yet they did not further improve 3-HP production in QLW58.

The dual nature of Uga1 for 3-HP production

The structural similarity between 3-HP and serine allowed us to identify the serine dehydrogenase Ora1 for 3-HP production (Fig. 2d), while the isomerized relation of lactate and 3-HP allowed us to identify the potential 3-HP permease Esb6p in yeast (Fig. 3b). Thus, we hypothesized that these similarities may also allow native enzymes to mediate the degradation of 3-HP or its intermediates (Fig. 6a). We started by deleting lactate dehydrogenases Dld1, Dld2, and Dld3, respectively, yet none of the deletions increased 3-HP production (Fig. 6b).

Moreover, it was demonstrated that the β-alanine-pyruvate aminotransferase (BAPAT) and gamma-aminobutyrate⁸² transaminase Uga1 exhibited substrate promiscuity and could convert β-alanine to MSA for 3-HP production⁸². The transamination reaction is highly reversible⁸³, while studies have also used BAPAT to convert MSA to β-alanine in the β-alanine auxotroph *E. coli*²⁰. These reports indicated that the Uga1 catalysed reaction is reversible, and it may also degrade MSA back to β-alanine, leading to energy waste and decreasing 3-HP yield (Fig. 6a). We therefore deleted *UGA1* in the strain and observed a slight increase in 3-HP production to 10.37 g/L in QLW65 (Fig. 6b). We also deleted alanine transaminases Alt1 and Alt2 that convert pyruvate to alanine, respectively, but it had no impact on the production of 3-HP (Fig. 6b, QLW66 and QLW67). The deletion of *UGA2* to prevent the potential degradation of MSA to malonate did not improve the production of 3-HP, either (Fig. 6b, QLW68). We believe that the *UGA1* deletion only slightly increased 3-HP production because the generation of MSA was restricted. Therefore, we used the strong promoters *CCW12* and *TDH3* to express the MCR-C domain protein (Fig. 6b). The strain with a strong *CCW12* promoter could increase 3-HP production to 10.77 g/L (Fig. 6b, QLW69), indicating that MSA production had been improved. We further deleted *UGA1* in this QLW69 strain and indeed increased 3-HP production to 11.25 g/L (Fig. 6b, QLW71). Meanwhile, we have tested the intracellular concentration of the proposed intermediate β-alanine, and indeed the deletion of *UGA1* in the

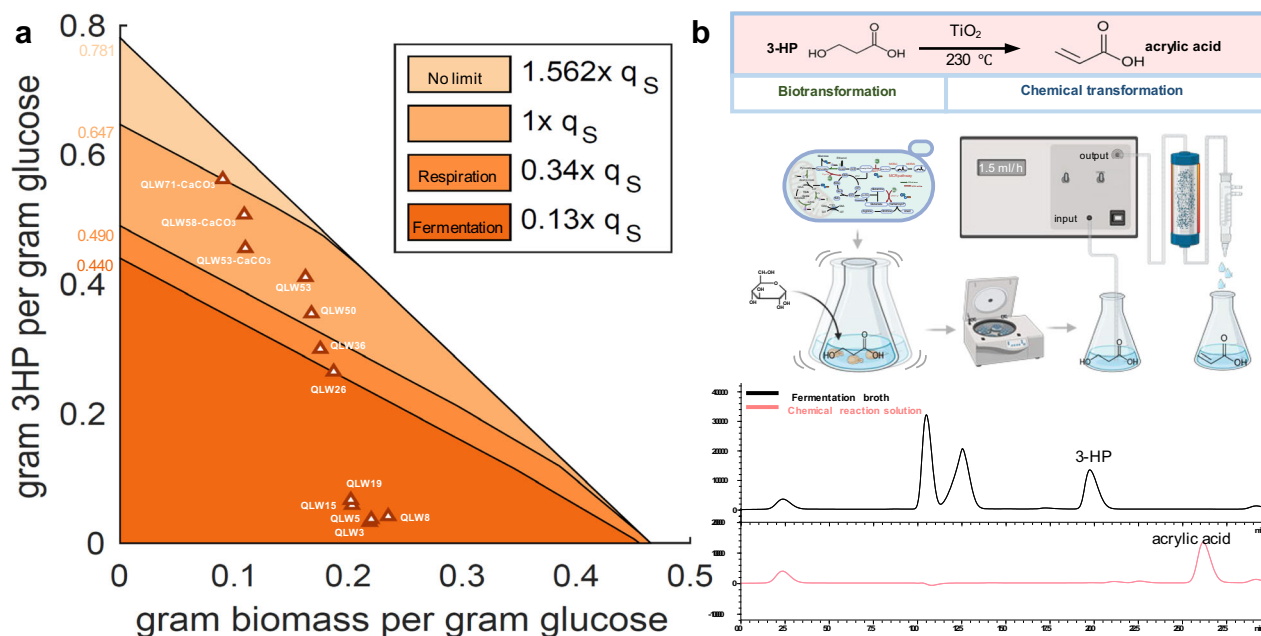


Fig. 7 | The final 3-HP production and its conversion to acrylic acid. **a** The production envelope incorporated with key 3-HP production results reported in this study. Collectively, four of our constructed strains exceeded the 3-HP production boundary with native metabolisms, with QLW71 approaching the

inaccessible one to one ratio of bicarbonate influx versus glucose uptake. q_S was substrate (glucose) consumption rate. **b** After a simple centrifugation step to remove cell pellets, the medium could be used directly for dehydration to generate acrylic acid. Source data are provided as a Source Data file.

Meanwhile, the production envelope analysis also revealed a carbon flux competition between the 3-HP production and the biomass generation. With the superposition of multiple strategies, the available carbon resources were insufficient to simultaneously accumulate high levels of both 3-HP and biomass. For example, when further rewiring the carbon flux towards 3-HP from QLW58 to QLW71, the strain could only move to the top left along the production envelope curve and result in less biomass generation (Fig. 7a). Importantly, the yield of QLW71 had reached the production envelope boundary of the bicarbonate supplementation equal to the glucose supplementation.

Through this significant rewiring of carbon flux, we achieved a high 3-HP yield. This high 3-HP titre significantly facilitated the downstream processing, that after simple centrifugation to remove cell pellets, the medium could be directly utilized for acrylate generation. Indeed, the 3-HP present in the medium underwent facile dehydration, facilitated by TiO_2 as a catalyst, ultimately yielding acrylate with a remarkable transformation ratio of 100% (Fig. 7b).

Discussion

Here we employed the production envelope analysis to demonstrate the crucial role of a high flux of bicarbonate in the production of the platform chemical 3-HP. We identified that the limited availability of bicarbonate has been a major obstacle in previous attempts to produce 3-HP (Fig. 1b), while our strategies of thorough analysis and rewiring of the native bicarbonate metabolism combined with external bicarbonate supply, enabled us to surpass this bicarbonate limitation for 3-HP production. Furthermore, to meet the energy demand for bicarbonate utilization, we employed the 5-InsP₇ mechanism to enhance the ATP generation yield. In addition, we implemented methods to reduce energy consumption, such as identifying an exporter for 3-HP to minimize energy waste during the 3-HP fermentation process. Moreover, we disrupted the degradation pathway of the intermediate metabolite MSA to prevent the occurrence of energy futile cycles. Collectively, we were able to achieve the in-flux of bicarbonate for 3-HP equivalent to that of glucose. Ultimately, in the QLW71 strain, the yield of 3-HP to substrate reached 0.5625 g/g, which corresponds to 89.3% of the theoretical yield with biomass generation. This

achievement surpassed the production envelope boundary of bicarbonate (Fig. 7a).

The pursuit of a circular and sustainable carbon economy is a key driver for development of biofuel and biobased chemical production from sustainable carbon sources. However, the net impact of biofuel use on global CO₂ emissions is still under debate^{84,85}, leading to continued iterative development⁸⁶. Carbon negativity strategies, such as capturing CO₂ or one-carbon feedstock into biomass or target molecules directly, have been proposed but require large energy inputs and may still result in CO₂ emissions^{87–89}. Our research focused on the high carbon yield production of 3-HP using a multi-strategy approach. Biobased commodity bulk chemicals require high feedstock and carbon yields to achieve economic efficacy and compete with petrochemical-based industries⁹⁰. The pathway for the production of these bulk chemicals must also decrease CO₂ release during glucose oxidation to have commercial potential. Our research provides a step towards the development of sustainable and circular carbon economies for biofuel and biobased chemical productions.

Methods

Strains, chemicals, and genetic manipulation

An *S. cerevisiae* strain CEN.PK 113-11 C (*MATa SUC2 MAL2-8⁺ his3Δ1 ura3-52*)⁵⁹ was used as the experimental model in this study. All the derivatives strains and their genotypes used in this study were listed in Supplementary Data 1 and 2. For the production of 3-HP, the defined minimal medium⁶⁰ was used. In detail, the medium consisted of 7.5 g/L (NH₄)₂SO₄, 14.4 g/L KH₂PO₄, and 0.5 g/L MgSO₄·7H₂O (the pH of the medium was adjusted to 6.5 with KOH). Trace metal and vitamin solutions were supplemented after the medium was autoclaved at 121 °C for 20 min. Histidine (60 mg/L) and uracil (60 mg/L) were supplemented based on the auxotroph of the plasmid. The calcium carbonate and the sodium bicarbonate were added if needed. To test the dry cell weight (DCW) after the fermentation, the 14.4 g/L KH₂PO₄ in the medium with the calcium carbonate fermentation was changed to 3.0 g/L KH₂PO₄ same as the medium reported for bioreactor cultivations⁹¹. For the medium with MES buffer, 100 mM MES monohydrate was added to the defined minimal medium till the pH was

adjusted to 6.5. The carbon source in all the media was 20 g/L glucose. The 3-HP standard was purchased from TCI (TCI, H0297). $\text{NaH}^{13}\text{CO}_3$ was purchased from Sigma-Aldrich (Sigma-Aldrich, 372382), 2-(*N*-morpholino) ethane sulfonic acid (MES) monohydrate was purchased from Sigma-Aldrich (Sigma-Aldrich, 69892), Nile red was purchased from (Thermo Fisher, N1142). Other chemicals including analytical standards were purchased from Sinopharm Chemical (Beijing, China). All genome modifications were carried out using the GTR-CRISPR system⁶¹. All plasmids were constructed by Golden Gate assembly. The MCR used in this study was from *Chloroflexus aurantiacus*²¹, phosphoketolase with substrate specificity to xylulose-5-phosphate was from *Leuconostoc mesenteroides*²¹, and phosphotransacetylase was from *Clostridium kluyveri*²². The gene of serine dehydrogenase *Oral* was amplified from the genome of the WT *S. cerevisiae* strain. The NADP⁺-dependent dehydrogenase *YdfG* was amplified from the genome of *E. coli* MG 1655. All genes were codon-optimized to *S. cerevisiae* and synthesized by Sangon Biotech (Shanghai, China) if not specified. Details of strains, and DNA fragments were summarized in Supplementary Data 3 and 4, respectively.

Growth analysis and shake flask fermentation

The strain used for 3-HP production was precultured in the defined minimal medium, then inoculated into a 100 mL shake flask containing 20 mL of the defined minimal medium with a final optical density at 600 nm (OD_{600}) of 0.1. The shake flask fermentation was carried out at 30 °C and 220 rpm for 72 h for 3-HP production. The samples' DCW was measured as reported⁷⁴. The biomass composition of $\text{CH}_{1.8}\text{O}_{0.5}\text{N}_{0.2}$ was assumed⁶⁵.

Analytical methods

Concentrations of 3-HP, glucose, and extracellular metabolites were determined using high-performance liquid chromatography (HPLC, Shimadzu, Japan) with an Aminex HPX-87H column (Bio-Rad, Hercules, USA) at 65 °C, using 0.5 mM H_2SO_4 as the mobile phase at 0.4 mL/min flow rate for 36 min⁷⁷. Glucose, glycerol, and acrylic acid concentration was detected using a RI-101 Refractive Index Detector, while 3-HP, OAA, acetate, and other extracellular metabolites were detected using a DAD-3000 Diode Array Detector at 210 nm.

3-HP tolerance test and intracellular 3-HP measurement

The minimal medium with 50 g/L 3-HP was used. 159 mL 0.5 M citrate solution and 41 mL 1 M Na_2HPO_4 were used to buffer the medium pH to 3.5. The pH of 3-HP standard liquid was buffered to neutral by NaOH and filtered to sterilize before added into the medium. For the growth test, the Growth Profiler 960 (EnzyScreen) was used. The initial inoculation OD_{600} was 0.01 and 200 μL of cells were cultivated in 96-well microplates at 30 °C and 220 rpm. The strain was cultured in a shake flask with the defined minimal medium and sampled at the exponential phase (OD_{600} around 1). Totally 40 OD_{600} cells were collected and resuspended into 2 mL test medium as an OD_{600} of 20. The test medium was same as the high 3-HP concentration growth test medium in Growth Profiler (pH = 3.5 and 50 g/L 3-HP was added). After be incubated at 30 °C and 220 rpm for 24 h, the culture was centrifuged for 2 min at 3000 $\times g$ and washed three times using ultra-pure water. The cells were then lysed by FastPrep-24 5 G for 20 s (6.0 m/s) with glass beads. After centrifuging for 1 min at 12,000 $\times g$, the supernatants, containing the cellular contents released upon lysis, were analysed by HPLC⁹².

Determination of β -alaine

After 72 h of fermentation in the shake flask with the defined minimal medium, 2 mL cell culture was centrifuged for 2 min at 3000 $\times g$ to collect the cells. The cell pellet was washed three times with sterile water, re-suspended in 1 mL sterile water and measured the cell OD_{600} . 400 μL glass beads were added to this cell suspension to get the cell

wall lysed by FastPrep-24 5 G for 20 s (6.0 m/s). The cell lysis was then centrifuged for 1 min at 12,000 $\times g$ and 10 μL supernatant was transferred into a glass HPLC vial for derivatization. Briefly, the AccQ•Fluor Kit (Waters, USA) was used for derivatization. 1 mL of acetonitrile was added to the AccQTag Ultra powder, mixed well, and heated to 55 °C to help it dissolve (heating time should not exceed 15 min). Then, 70 μL of borate buffer was added to the glass HPLC vial containing 10 μL of the cell lysis solution, as well as 20 μL of prepared AccQTag Ultra derivatization solution, mix well, and heated at 55 °C for 10 min. Finally, the heated sample was diluted with derivatization diluent at a ratio of 1:100. The diluted sample was then analyzed using UHPLC-MS/MS. The analysis was performed using Acquity UPLC binary solvent manager, autosampler manager, and column manager (Waters, Milford, MA, USA) coupled with Xevo TQ-S tandem quadrupole mass spectrometer (Waters, Wilmslow, U.K.). Reverse-phase gradient chromatography was employed using an HSS T3 2.1 \times 150 mm, 1.8 μm column. The mobile phase A was 0.1% formic acid in water (v/v), while mobile phase B was 1% formic acid in acetonitrile (v/v). The column temperature was maintained at 45 °C. The elution was performed at a flow rate of 0.6 mL/min with a linear gradient: started at 4% B for 0.5 min, increased to 10% B at 2 min, then 28% B at 2.5 min, finally to 95% B for 1 min, and then returned to 4% B for column re-equilibration⁹³.

Gene expression analysis

Samples for quantitative real-time PCR (qRT-PCR) analysis were taken when the OD_{600} of the cells reached around 1. The cell pellet was collected by centrifugation at -20 °C, quenched by liquid nitrogen, and stored at -80 °C before use. The RNeasy Mini Kit (QIAGEN, USA) was used for the extraction of total RNA as recommended. The QuantiTect Reverse Transcription Kit (QIAGEN, USA) was used for reverse transcription. 2 μL of produced cDNA was used as the template of the qPCR reaction with the DyNAmo Flash SYBR Green qPCR Kit (Thermo Scientific, USA). Quantitative RT-PCR was performed on Stratagene Mx3005P (Agilent Technologies, USA). The housekeeping gene *ACT1* was selected as the reference gene. All the primers were listed in Supplementary Data 4.

Catalytic dehydration of 3-HP to acrylic acid

Acrylic acid synthesis was carried out in a fixed-bed tubular reactor (i.d. = 8 mm, o.d. = 12 mm, length = 30 cm). After culturing the strains for 72 h in a defined minimal medium, the fermentation broth was sterilized by filtration and diluted 6 times with ultrapure water. And then 3 g TiO_2 as the catalyst (20–40 mesh) was loaded into the reactor tube between two layers of quartz wool and preheated at 230 °C in the furnace. The diluted fermentation broth was delivered using a metering pump with a rate of 100 mL/h for 5 min, and then with a constant rate of 1.5 mL/h. The eluate containing acrylic acid was condensed in a cold-water bath. Liquid samples of the first 30 min were discarded. In addition, 4 samples were collected for each measurement.

Genome-scale metabolic modelling

The genome-scale metabolic model of *S. cerevisiae* Yeast8¹⁷, was expanded by adding two heterologous reactions, i.e., malonyl-CoA reduction and MSA reduction, as well as the secretion of 3-HP. The resulting model with the default constraints was used to estimate the yield of 3-HP on glucose. To do so, the uptake rate of glucose was fixed at 1 mmol/gDCW/h, the growth rate was adjusted, and the secretion of 3-HP was maximized for each growth rate. To simulate the effect of native bicarbonate production on the yield, the upper bound on the rate of the bicarbonate formation in the model, which converts carbon dioxide to bicarbonate, was set at different values normalized to the glucose uptake rate. Model simulations were performed using the COBRA toolbox⁹⁴ in MATLAB with the CPLEX as an optimization solver.

Phylogenetic and structure analysis

Amino acid sequences from the SLC4 family (InterPro ID: IPR003020) and SLC26 family (InterPro ID: IPR001902) were aligned and trimmed using NGPhylogeny.fr⁹⁵. Only key species (*S. cerevisiae*, *Arabidopsis thaliana*, and *Homo sapiens* for SLC4 and *S. cerevisiae*, *E. coli*, *A. thaliana*, and *H. sapiens* for SLC26) were included. MEGA11 was used to construct a maximum likelihood phylogenetic tree and tested with 500 bootstraps⁹⁶. iTOL v6 was used in the modification of the phylogenetic tree⁹⁷. The structure of human SLC26A11 (Uniprot ID Q86WA9), yeast Sul1 (Uniprot ID P38359), and yeast Sul2 (Uniprot ID Q12325) were obtained from AlphaFoldDB (<https://alphafold.ebi.ac.uk/>)⁹⁸. FATCAT 2.0 (<https://fatcat.godziklab.org/>)⁹⁹ was used to compare the structures of SLC26A11 with Sul1 and Sul2.

Isotope ¹³C labelling analysis

The minimal medium with 30 mM NaH¹³CO₃ (Sigma-Aldrich) in a shake flask was used for the isotope labelling experiment. Strains were inoculated with OD₆₀₀ 0.01 and sampled after 72 h. 2 mL medium was transferred immediately into the 15 mL tube with 8 mL pure methanol (precooling in -80 °C fridge) and vortexed around 1 s for quenching samples. Then, the mix was centrifuged at 5000 × g for 3 min at 4 °C. Remove the supernatant and resuspend the precipitate with 3 mL of precooling methanol, 5000 × g centrifuged for 3 min, remove the supernatant, keep on ice. For hydrolysis of cellular components, 6 mL 6 N hydrochloric acid was added to quenched cells and transferred to 10 mL glass digestion tube, cells were hydrolysed for 16–20 h in metal bath at 115 °C and dried in a heating block at 85 °C, then 600 µL of ultrapure water was added, and the sample was centrifuged at the maximum speed for 1 min¹⁰⁰. Before the MS analysis, the sample was filtrated with a 0.22 µm pore size filter membrane. The isotopic labels of metabolites were determined with UPLC-MS/MS (SCIEX TRIPLETOF 6600).

Nile red staining

After 96 h of cultivation in the defined minimal medium, 250 µL cell culture was transferred into a sterile 1.5 mL tube. Then 25 µL freshly prepared DMSO: PBS (1:1) buffer and 25 µL Nile red (60 µg/mL in acetone) were added. The working concentration of Nile red was 5 µg/mL. The sample was well mixed and then incubated away from the light for 5 min at room temperature. The strain was washed twice with 300 µL PBS buffer and resuspended in 300 µL PBS buffer. 5 µL of the cell sample was pipetted onto a glass microscope slide, placed with the cover slip firmly, and observed using the green fluorescence channel with the excitation wavelength at 488 nm and emission wavelength at 509 nm. For quantitative analysis, the yeast culture in the defined minimal medium was harvested at 96 h, centrifuged and resuspended twice with PBS buffer to achieve an OD₆₀₀ of 2.0. Transfer 250 µL of each yeast sample to the corresponding wells of a 96-well plate, ensuring proper mixing as yeast cells tend to settle. 12.5 µL DMSO: PBS (1:1) buffer for each well was added in advance. 25 µL of the freshly prepared 60 µg/mL Nile red stock solution was added for each well to a final 5 µg/mL concentration. The microplate reader with excitation and emission wavelengths was set at 485 nm and 535 nm, respectively, and a top 50% mirror was used. Optimal gain settings and an orbital shake for 10 s before reading were employed and the plate was read immediately after Nile red addition for the necessary number of cycles¹⁰¹.

Reporting summary

Further information on research design is available in the Nature Portfolio Reporting Summary linked to this article.

Data availability

The data supporting the findings of this work are available within the paper and the Supplementary Information files. A reporting summary

for this article is available as a Supplementary Information file. Source data are provided with this paper.

References

1. Scown, C. D. & Keasling, J. D. Sustainable manufacturing with synthetic biology. *Nat. Biotechnol.* **40**, 304–307 (2022).
2. Keasling, J. et al. Microbial production of advanced biofuels. *Nat. Rev. Microbiol.* **19**, 701–715 (2021).
3. Liu, Y. et al. Biofuels for a sustainable future. *Cell* **184**, 1636–1647 (2021).
4. Hertel, T. W. et al. Effects of us maize ethanol on global land use and greenhouse gas emissions: estimating market-mediated responses. *Bioscience* **60**, 223–231 (2010).
5. Liu, Z., Wang, K., Chen, Y., Tan, T. & Nielsen, J. Third-generation biorefineries as the means to produce fuels and chemicals from CO₂. *Nat. Catal.* **3**, 274–CO288 (2020).
6. Baumschabl, M. et al. Conversion of CO₂ into organic acids by engineered autotrophic yeast. *Proc. Natl Acad. Sci. USA* **119**, e2211827119 (2022).
7. Fuchs, G. Alternative pathways of carbon dioxide fixation: insights into the early evolution of life? *Annu. Rev. Microbiol.* **65**, 631–658 (2011).
8. Bianchi, F., Van't Klooster, J. S., Ruiz, S. J. & Poolman, B. Regulation of amino acid transport in *Saccharomyces cerevisiae*. *Microbiol. Mol. Biol. Rev.* **83**, 10–1128 (2019).
9. Berg, I. A. Ecological aspects of the distribution of different autotrophic CO₂ fixation pathways. *Appl. Environ. Microbiol.* **77**, 1925–1936 (2011).
10. Berg, I. A., Kockelkorn, D., Buckel, W. & Fuchs, G. A 3-hydroxypropionate/4-hydroxybutyrate autotrophic carbon dioxide assimilation pathway in Archaea. *Science* **318**, 1782–1786 (2007).
11. Zarzycki, J., Brecht, V., Muller, M. & Fuchs, G. Identifying the missing steps of the autotrophic 3-hydroxypropionate CO₂ fixation cycle in *Chloroflexus aurantiacus*. *Proc. Natl Acad. Sci. USA* **106**, 21317–21322 (2009).
12. Werpy, T. & Petersen, G. Top value added chemicals from biomass volume I—Results of screening for potential candidates from sugars and synthesis gas. No. DOE/GO-102004-1992. National Renewable Energy Lab. (NREL), Golden, CO (United States) (2004).
13. Zabed, H. M. et al. Biocatalytic gateway to convert glycerol into 3-hydroxypropionic acid in waste-based biorefineries: Fundamentals, limitations, and potential research strategies. *Biotechnol. Adv.* **62**, 108075 (2023).
14. Dishisha, T., Pyo, S. H. & Hatti-Kaul, R. Bio-based 3-hydroxypropionic- and acrylic acid production from biodiesel glycerol via integrated microbial and chemical catalysis. *Microb. Cell Fact.* **14**, 200 (2015).
15. Bhagwat, S. S. et al. Sustainable production of acrylic acid via 3-hydroxypropionic acid from lignocellulosic biomass. *ACS Sustain. Chem. Eng.* **9**, 16659–16669 (2021).
16. Yu, W., Cao, X., Gao, J. & Zhou, Y. J. Overproduction of 3-hydroxypropionate in a super yeast chassis. *Bioresour. Technol.* **361**, 127690 (2022).
17. Lu, H. et al. A consensus *S. cerevisiae* metabolic model Yeast8 and its ecosystem for comprehensively probing cellular metabolism. *Nat. Commun.* **10**, 3589 (2019).
18. Canelas, A. B. et al. Integrated multilaboratory systems biology reveals differences in protein metabolism between two reference yeast strains. *Nat. Commun.* **1**, 145 (2010).
19. Hugler, M., Menendez, C., Schagger, H. & Fuchs, G. Malonyl-coenzyme A reductase from *Chloroflexus aurantiacus*, a key enzyme of the 3-hydroxypropionate cycle for autotrophic CO₂ fixation. *J. Bacteriol.* **184**, 2404–2410 (2002).

20. Liu, C. et al. Functional balance between enzymes in malonyl-CoA pathway for 3-hydroxypropionate biosynthesis. *Metab. Eng.* **34**, 104–111 (2016).
21. Qin, N. et al. Rewiring central carbon metabolism ensures increased provision of acetyl-CoA and NADPH required for 3-OH-propionic acid production. *ACS Synth. Biol.* **9**, 3236–3244 (2020).
22. Chen, X., Yang, X., Shen, Y., Hou, J. & Bao, X. Increasing malonyl-CoA derived product through controlling the transcription regulators of phospholipid synthesis in *Saccharomyces cerevisiae*. *ACS Synth. Biol.* **6**, 905–912 (2017).
23. Ouyang, L., Holland, P., Lu, H., Bergenholm, D. & Nielsen, J. Integrated analysis of the yeast NADPH-regulator Stb5 reveals distinct differences in NADPH requirements and regulation in different states of yeast metabolism. *FEMS Yeast Res.* **18**, foy091 (2018).
24. Christodoulou, D. et al. Reserve Flux Capacity in the pentose phosphate pathway enables *Escherichia coli*'s rapid response to oxidative stress. *Cell Syst.* **6**, 569–578.e567 (2018).
25. Kim, J. E., Jang, I. S., Sung, B. H., Kim, S. C. & Lee, J. Y. Rerouting of NADPH synthetic pathways for increased protopanaxadiol production in *Saccharomyces cerevisiae*. *Sci. Rep.* **8**, 15820 (2018).
26. Curran, K. A. et al. Design of synthetic yeast promoters via tuning of nucleosome architecture. *Nat. Commun.* **5**, 4002 (2014).
27. Xiao, M., Zhu, X., Bi, C., Ma, Y. & Zhang, X. Improving succinate productivity by engineering a cyanobacterial CO₂ concentrating system (CCM) in *Escherichia coli*. *Biotechnol. J.* **12**, 1700199 (2017).
28. Yu, J. H. et al. Combinatorial optimization of CO₂ transport and fixation to improve succinate production by promoter engineering. *Biotechnol. Bioeng.* **113**, 1531–1541 (2016).
29. Aguilera, J., Van Dijken, J. P., De Winde, J. H. & Pronk, J. T. Carbonic anhydrase (Nce103p): an essential biosynthetic enzyme for growth of *Saccharomyces cerevisiae* at atmospheric carbon dioxide pressure. *Biochem. J.* **391**, 311–316 (2005).
30. Cordat, E. & Reithmeier, R. A. Structure, function, and trafficking of SLC4 and SLC26 anion transporters. *Curr. Top. Membr.* **73**, 1–67 (2014).
31. Jennings, M. L., Howren, T. R., Cui, J., Winters, M. & Hannigan, R. Transport and regulatory characteristics of the yeast bicarbonate transporter homolog Bor1p. *Am. J. Physiol. Cell Physiol.* **293**, C468–C476 (2007).
32. Alper, S. L. & Sharma, A. K. The SLC26 gene family of anion transporters and channels. *Mol. Asp. Med.* **34**, 494–515 (2013).
33. Cherest, H. et al. Molecular characterization of two high affinity sulfate transporters in *Saccharomyces cerevisiae*. *Genetics* **145**, 627–635 (1997).
34. Soleimani, M. SLC26 Cl⁻/HCO₃⁻ exchangers in the kidney: roles in health and disease. *Kidney Int.* **84**, 657–666 (2013).
35. Guo, X. et al. Ptc7p dephosphorylates select mitochondrial proteins to enhance metabolic function. *Cell Rep.* **18**, 307–313 (2017).
36. Guo, X., Niemi, N. M., Coon, J. J. & Pagliarini, D. J. Integrative proteomics and biochemical analyses define Ptc6p as the *Saccharomyces cerevisiae* pyruvate dehydrogenase phosphatase. *J. Biol. Chem.* **292**, 11751–11759 (2017).
37. Yu, T. et al. Reprogramming yeast metabolism from alcoholic fermentation to lipogenesis. *Cell* **174**, 1549–1558.e1514 (2018).
38. Usaite, R. et al. Reconstruction of the yeast Snf1 kinase regulatory network reveals its role as a global energy regulator. *Mol. Syst. Biol.* **5**, 319 (2009).
39. Wang, S. et al. The expression modulation of the key enzyme acc for highly efficient 3-hydroxypropionic acid production. *Front. Microbiol.* **13**, 902848 (2022).
40. Shi, S., Chen, Y., Siewers, V. & Nielsen, J. Improving production of malonyl coenzyme A-derived metabolites by abolishing Snf1-dependent regulation of Acc1. *MBio* **5**, e01130–01114 (2014).
41. Son, H. F. et al. Structural insight into bi-functional malonyl-CoA reductase. *Environ. Microbiol.* **22**, 752–765 (2020).
42. Fujisawa, H., Nagata, S. & Misono, H. Characterization of short-chain dehydrogenase/reductase homologues of *Escherichia coli* (YdfG) and *Saccharomyces cerevisiae* (YMR226C). *Biochim. Biophys. Acta* **1645**, 89–94 (2003).
43. Zhu, Z. et al. Expanding the product portfolio of fungal type I fatty acid synthases. *Nat. Chem. Biol.* **13**, 360–362 (2017).
44. Tong, T. et al. A biosynthesis pathway for 3-hydroxypropionic acid production in genetically engineered *Saccharomyces cerevisiae*. *Green. Chem.* **23**, 4502–4509 (2021).
45. Qin, N., Li, L., Wang, Z. & Shi, S. Microbial production of odd-chain fatty acids. *Biotechnol. Bioeng.* **120**, 917–931 (2022).
46. Keren, L. et al. Promoters maintain their relative activity levels under different growth conditions. *Mol. Syst. Biol.* **9**, 701 (2013).
47. Hiltunen, J. K. et al. The biochemistry of peroxisomal beta-oxidation in the yeast *Saccharomyces cerevisiae*. *FEMS Microbiol. Rev.* **27**, 35–64 (2003).
48. Peetermans, A., Foulque-Moreno, M. R. & Thevelein, J. M. Mechanisms underlying lactic acid tolerance and its influence on lactic acid production in *Saccharomyces cerevisiae*. *Microb. Cell* **8**, 111–130 (2021).
49. Kildegaard, K. R. et al. Evolution reveals a glutathione-dependent mechanism of 3-hydroxypropionic acid tolerance. *Metab. Eng.* **26**, 57–66 (2014).
50. Xu, X., Williams, T. C., Divne, C., Pretorius, I. S. & Paulsen, I. T. Evolutionary engineering in *Saccharomyces cerevisiae* reveals a Trk1-dependent potassium influx mechanism for propionic acid tolerance. *Biotechnol. Biofuels* **12**, 97 (2019).
51. Chun, A. Y., Yunxiao, L., Ashok, S., Seol, E. & Park, S. Elucidation of toxicity of organic acids inhibiting growth of *Escherichia coli* W. *Biotech. and Biopro. Eng* **19**, 858–865 (2014).
52. Van Maris, A. J., Konings, W. N., van Dijken, J. P. & Pronk, J. T. Microbial export of lactic and 3-hydroxypropanoic acid: implications for industrial fermentation processes. *Metab. Eng.* **6**, 245–255 (2004).
53. Lee, J. J., Crook, N., Sun, J. & Alper, H. S. Improvement of lactic acid production in *Saccharomyces cerevisiae* by a deletion of Ssb1. *J. Ind. Microbiol. Biotechnol.* **43**, 87–96 (2016).
54. Baek, S. H., Kwon, E. Y., Kim, S. Y. & Hahn, J. S. GSF2 deletion increases lactic acid production by alleviating glucose repression in *Saccharomyces cerevisiae*. *Sci. Rep.* **6**, 34812 (2016).
55. Sherwood, P. W. & Carlson, M. Efficient export of the glucose transporter Hxt1p from the endoplasmic reticulum requires Gsf2p. *Proc. Natl Acad. Sci. USA* **96**, 7415–7420 (1999).
56. Darbani, B., Stovicek, V., van der Hoek, S. A. & Borodina, I. Engineering energetically efficient transport of dicarboxylic acids in yeast *Saccharomyces cerevisiae*. *Proc. Natl Acad. Sci. USA* **116**, 19415–19420 (2019).
57. Baek, S. H. et al. Improvement of d-lactic acid production in *Saccharomyces cerevisiae* under acidic conditions by evolutionary and rational metabolic engineering. *Biotechnol. J.* **12**, 1700015 (2017).
58. Dato, L. et al. Changes in SAM2 expression affect lactic acid tolerance and lactic acid production in *Saccharomyces cerevisiae*. *Microb. Cell Fact.* **13**, 147 (2014).
59. Nygard, Y. et al. The diverse role of Pdr12 in resistance to weak organic acids. *Yeast* **31**, 219–232 (2014).
60. Viegas, C. A., Sebastiao, P. B., Nunes, A. G. & Sa-Correia, I. Activation of plasma membrane H(+)-ATPase and expression of PMA1 and PMA2 genes in *Saccharomyces cerevisiae* cells grown at supraoptimal temperatures. *Appl. Environ. Microbiol.* **61**, 1904–1909 (1995).
61. Orij, R., Brul, S. & Smits, G. J. Intracellular pH is a tightly controlled signal in yeast. *Biochem. Biophys. Acta* **1810**, 933–944 (2011).
62. Sugiyama, M., Akase, S. P., Nakanishi, R., Kaneko, Y. & Harashima, S. Overexpression of ESBP6 improves lactic acid resistance and

- production in *Saccharomyces cerevisiae*. *J. Biosci. Bioeng.* **122**, 415–420 (2016).
63. Casal, M., Paiva, S., Andrade, R. P., Gancedo, C. & Leão, C. L. The lactate-proton symport of *Saccharomyces cerevisiae* is encoded by *JEN1*. *J. Bacteriol.* **181**, 2620–2623 (1999).
64. Pacheco, A. et al. Lactic acid production in *Saccharomyces cerevisiae* is modulated by expression of the monocarboxylate transporters *Jen1* and *Ady2*. *FEMS Yeast Res.* **12**, 375–381 (2012).
65. Baek, S. H., Kwon, E. Y., Kim, Y. H. & Hahn, J. S. Metabolic engineering and adaptive evolution for efficient production of D-lactic acid in *Saccharomyces cerevisiae*. *Appl. Microbiol. Biotechnol.* **100**, 2737–2748 (2016).
66. Serre, V. et al. Half of *Saccharomyces cerevisiae* carbamoyl phosphate synthetase produces and channels carbamoyl phosphate to the fused aspartate transcarbamoylase domain. *J. Biol. Chem.* **274**, 23794–23801 (1999).
67. Ali, E. S. et al. The mTORC1-SLC4A7 axis stimulates bicarbonate import to enhance de novo nucleotide synthesis. *Mol. Cell* **82**, 3284–3298.e7 (2022).
68. Winkler, M. E. & Ramos-Montanez, S. Biosynthesis of histidine. *EcoSal. Plus* **3**, 10–1128 (2009).
69. Alam, M. T. et al. The metabolic background is a global player in *Saccharomyces* gene expression epistasis. *Nat. Microbiol.* **1**, 15030 (2016).
70. Yan, C. et al. Auxotrophs compromise cell growth and fatty acid production in *Saccharomyces cerevisiae*. *Biotechnol. J.* **18**, e2200510 (2023).
71. Yu, R. et al. Nitrogen limitation reveals large reserves in metabolic and translational capacities of yeast. *Nat. Commun.* **11**, 1881 (2020).
72. Malubhoy, Z. et al. Carbon dioxide fixation via production of succinic acid from glycerol in engineered *Saccharomyces cerevisiae*. *Microb. Cell Fact.* **21**, 102 (2022).
73. Lopez Garzon, C. S. & Straathof, A. J. Recovery of carboxylic acids produced by fermentation. *Biotechnol. Adv.* **32**, 873–904 (2014).
74. Rodriguez, A., De La Cera, T., Herrero, P. & Moreno, F. The hexokinase 2 protein regulates the expression of the *GLK1*, *HXK1* and *HXK2* genes of *Saccharomyces cerevisiae*. *Biochem. J.* **355**, 625–631 (2001).
75. Stoddard, P. R. et al. Polymerization in the actin ATPase clan regulates hexokinase activity in yeast. *Science* **367**, 1039–1042 (2020).
76. Bakker, B. M. et al. Stoichiometry and compartmentation of NADH metabolism in *Saccharomyces cerevisiae*. *FEMS Microbiol. Rev.* **25**, 15–37 (2001).
77. Moreno, F. & Herrero, P. The hexokinase 2-dependent glucose signal transduction pathway of *Saccharomyces cerevisiae*. *FEMS Microbiol. Rev.* **26**, 83–90 (2002).
78. Qin, N. et al. Flux regulation through glycolysis and respiration is balanced by inositol pyrophosphates in yeast. *Cell* **186**, 748–763 (2023).
79. Dai, Z., Huang, M., Chen, Y., Siewers, V. & Nielsen, J. Global rewiring of cellular metabolism renders *Saccharomyces cerevisiae* Crabtree negative. *Nat. Commun.* **9**, 3059 (2018).
80. Steidle, E. A. et al. A novel inositol pyrophosphate phosphatase in *Saccharomyces cerevisiae*. *J. Biol. Chem.* **291**, 6772–6783 (2016).
81. Szigyarto, Z., Garedew, A., Azevedo, C. & Saiardi, A. Influence of inositol pyrophosphates on cellular energy dynamics. *Science* **334**, 802–805 (2011).
82. Borodina, I. et al. Establishing a synthetic pathway for high-level production of 3-hydroxypropionic acid in *Saccharomyces cerevisiae* via beta-alanine. *Metab. Eng.* **27**, 57–64 (2015).
83. Litwack, G. *Human Biochemistry* 403–440 (Elsevier Inc, 2022).
84. Müller, L. J. et al. A Guideline for life cycle assessment of carbon capture and utilization. *Front. Energy Res.* **8**, 15 (2020).
85. DeCicco, J. M. et al. Carbon balance effects of U.S. biofuel production and use. *Clim. Change* **138**, 667–680 (2016).
86. Marcellin, E. et al. Low carbon fuels and commodity chemicals from waste gases – systematic approach to understand energy metabolism in a model acetogen. *Green. Chem.* **18**, 3020–3028 (2016).
87. Gleizer, S. et al. Conversion of *Escherichia coli* to generate all biomass carbon from CO₂. *Cell* **179**, 1255–1263.e1212 (2019).
88. Chen, F. Y., Jung, H. W., Tsuei, C. Y. & Liao, J. C. Converting *Escherichia coli* to a synthetic methylotroph growing solely on methanol. *Cell* **182**, 933–946.e914 (2020).
89. Antonovsky, N. et al. Sugar synthesis from CO₂ in *Escherichia coli*. *Cell* **166**, 115–125 (2016).
90. Hu, G. et al. Light-driven CO₂ sequestration in *Escherichia coli* to achieve theoretical yield of chemicals. *Nat. Catal.* **4**, 395–406 (2021).
91. Liu, Q. et al. Rewiring carbon metabolism in yeast for high level production of aromatic chemicals. *Nat. Commun.* **10**, 4976 (2019).
92. Li, C. et al. The eisosomes contribute to acid tolerance of yeast by maintaining cell membrane integrity. *Food Microbiol.* **110**, 104157 (2023).
93. Gray, N. et al. High-speed quantitative UPLC-MS analysis of multiple amines in human plasma and serum via precolumn derivatization with 6-aminoquinolyl-N-hydroxysuccinimidyl carbamate: application to acetaminophen-induced liver failure. *Anal. Chem.* **89**, 2478–2487 (2017).
94. Heirendt, L. et al. Creation and analysis of biochemical constraint-based models using the COBRA Toolbox v.3.0. *Nat. Protoc.* **14**, 639–702 (2019).
95. Lemoine, F. et al. NGPhylogeny.fr: new generation phylogenetic services for non-specialists. *Nucleic Acids Res.* **47**, W260–W265 (2019).
96. Tamura, K., Stecher, G. & Kumar, S. MEGA11: Molecular evolutionary genetics analysis version 11. *Mol. Biol. Evol.* **38**, 3022–3027 (2021).
97. Letunic, I. & Bork, P. Interactive Tree Of Life (iTOL) v5: an online tool for phylogenetic tree display and annotation. *Nucleic Acids Res.* **49**, W293–W296 (2021).
98. Jumper, J. et al. Highly accurate protein structure prediction with AlphaFold. *Nature* **596**, 583–589 (2021).
99. Li, Z., Jaroszewski, L., Iyer, M., Sedova, M. & Godzik, A. FATCAT 2.0: towards a better understanding of the structural diversity of proteins. *Nucleic Acids Res.* **48**, W60–W64 (2020).
100. Canelas, A. B. et al. Quantitative evaluation of intracellular metabolite extraction techniques for yeast metabolomics. *Anal. Chem.* **81**, 7379–7389 (2009).
101. Rostron, K. A. & Lawrence, C. L. Nile red staining of neutral lipids in yeast. *Methods Mol. Biol.* **1560**, 219–229 (2017).
102. Li, S., Si, T., Wang, M. & Zhao, H. Development of a synthetic malonyl-CoA sensor in *Saccharomyces cerevisiae* for intracellular metabolite monitoring and genetic screening. *ACS Synth. Biol.* **4**, 1308–1315 (2015).
103. David, F., Nielsen, J. & Siewers, V. Flux control at the malonyl-CoA node through hierarchical dynamic pathway regulation in *Saccharomyces cerevisiae*. *ACS Synth. Biol.* **5**, 224–233 (2016).
104. Zhang, Y., Su, M., Wang, Z., Nielsen, J. & Liu, Z. Rewiring regulation on respiro-fermentative metabolism relieved Crabtree effects in *Saccharomyces cerevisiae*. *Synth. Syst. Biotech.* **7**, 1034–1043 (2022).
105. Kildegaard, K. R. et al. Engineering and systems-level analysis of *Saccharomyces cerevisiae* for production of 3-hydroxypropionic acid via malonyl-CoA reductase-dependent pathway. *Microb. Cell Fact.* **15**, 53 (2016).
106. Zhang, Y. et al. Engineering yeast mitochondrial metabolism for 3-hydroxypropionate production. *Biotech. Biofuels. Bioprod.* **16**, 1–11 (2023).

Acknowledgements

This work was supported by National Key Research and Development Program of China (2018YFA0900100), National Natural Science Foundation of China (22078012), Fundamental Research Funds for the Central Universities (buctrc202304), the Beijing Advanced Innovation Centre for Soft Matter Science and Engineering, Beijing University of Chemical Technology, the Novo Nordisk Foundation (NNF20CC0035580) and the Swedish Research Council FORMAS. This paper was edited using ChatGPT. We also thank Professor Jiqin Zhu from Beijing University of Chemical Technology for his kind help on the acrylic acid experiment.

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Competing interests

The authors declare no competing interests.

Additional information

Supplementary information The online version contains supplementary material available at <https://doi.org/10.1038/s41467-024-45557-9>.

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Peer review information *Nature Communications* thanks Ji-Sook Hahn and the other, anonymous, reviewer(s) for their contribution to the peer review of this work. A peer review file is available.

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