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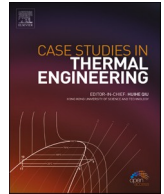
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# Case Studies in Thermal Engineering

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## Drying of cable backfill by vapour diffusion balanced by capillary water suction: a laboratory experiment with altered groundwater levels

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### ABSTRACT

The performance of underground infrastructures, such as an electrical transmission cable, is strongly affected by transmission-induced high temperatures, which have an impact on the properties of the cable and the electric load. Therefore, it is essential to determine if a high load will cause a dry out of a sandy thermal backfill close to the cable. This paper aims at describing processes and establishing criteria for when a dry out can occur. Processes that are addressed are e.g., vapour diffusion and capillary water transport. Criteria of interest are e.g., groundwater levels and capillarity of soil. A new pilot laboratory experiment, with varying groundwater levels, was developed to investigate the dry out of the cable sand in close proximity to an operating cable. The experimental results show that a thermal backfill can withstand a dry out, at a high current load in a cable, if there is a capillary contact to a groundwater level 2 m below the backfill material.

### Symbols

$T$	Temperature (K)
$\rho c$	Volumetric heat capacity ( $J/m^3K$ )
$\dot{q}$	Heat source ( $W/m^3$ )
$Q$	Heat loss (W)
$\lambda$	Thermal conductivity ( $W/mK$ )
$\emptyset$	Diameter (mm)
$t$	Time (s)
$d_{50}$	Medium value of grain size distribution (mm)
$m$	Mass (kg)
$\rho$	Density ( $kg/m^3$ )
$C_p$	Specific heat capacity ( $J/Kg/K$ )
$V$	Volume ( $m^3$ )
$A$	Surface area ( $m^2$ )
$Ke$	Kersten number (–)

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$h$  Heat transfer coefficient at external surfaces ( $\text{W}/\text{m}^2\text{K}$ )

#### Subscripts

*app* apparent  
*ac* after compaction  
*as* after saturation  
*as* after saturation  
*ee* end of experiment  
*c* convection  
*s* surface  
*w* water  
 dry dried at  $105\text{ }^\circ\text{C}$   
*a* air  
 sat saturated

#### Acronyms

HVDC High Voltage Direct Current  
 H cable sand Hamneda  
 SS cable sand Södra Sandby  
 GWC Gravimetric Water Content ( $\text{Kg}/\text{kg}$ )  
 VWC Volumetric Water Content ( $\text{m}^3/\text{m}^3$ )

## 1. Introduction

Underground high voltage direct current (HVDC) cables have the advantage of taking up less surface space. They are also less affected by surface climate conditions than traditional overhead transmission lines, which also have a negative esthetical impact [1]. In an operational HVDC cable, heat is generated due to the electric resistance of the conductor, a resistance that increases with increasing temperatures. The current-carrying capacity (the ampacity) of a cable system is influenced by the ability of the installation to dissipate the heat from the cable, into the surrounding medium [2,3]. The maximum operating temperature of the conductor varies between  $55$  and  $90\text{ }^\circ\text{C}$  among different cable types [4], where a conductor temperature of  $90\text{ }^\circ\text{C}$  approximately leads to a cable surface temperature of  $70\text{ }^\circ\text{C}$  [5]. A cable is designed for an ampacity that generates a certain maximum temperature, related to the lowest allowed thermal conductivity ( $\lambda$ ) of the thermal backfill (cable sand), which is surrounding the cable. Thus, a highly conductive backfill environment would allow for higher ampacity. However, if the design temperature is exceeded, it may influence the warranty from the manufacturer [1]. The functionality and lifetime of the cable can, in extreme cases, lead to a thermal runaway, and breakdown of the cable [1,6].

Thermal energy can be transported in soil by heat conduction, radiation, and convection. Radiation generally has minor influence [7] but can be of importance in e.g., very coarse material with low water content at elevated temperature. At higher temperatures, heat can also be transferred by diffusion of water in both the gas and liquid phase. In addition, advective heat transport can also take place by forced convection e.g., groundwater flow driven by hydraulic gradients [8]. Above the groundwater level, and at normal ground temperatures, heat conduction is the governing transport mechanism. The heat flow is linearly dependent on  $\lambda$ . High water content, high density of soil, and high quartz content are the most important factors to obtain high  $\lambda$  at normal temperatures. To determine  $\lambda$  of a material at elevated temperature or experimentally in the field, it is appropriate to use apparent thermal conductivity ( $\lambda_{app}$ ), since other heat transfer mechanisms may contribute to the result. At higher temperatures and intermediate degrees of saturation, vapour diffusion contributes to raising  $\lambda_{app}$  and increasing the heat transfer.

Near the cable surface, evaporation of water takes place and vapour is transported in the direction of the heat flow. This is illustrated in Fig. 1, where vapour migration induces a lower water content near the cable, which in turn induces a capillary water suction in the opposite direction, towards the cable. However, if the suction of liquid water for some reason cannot balance the vapour diffusion, the water content decreases, which leads to a decrease in  $\lambda_{app}$  and elevated temperature.

The heat flow, induced by the cable, is the main driving force for all other mechanisms. A low cable temperature gives less influence

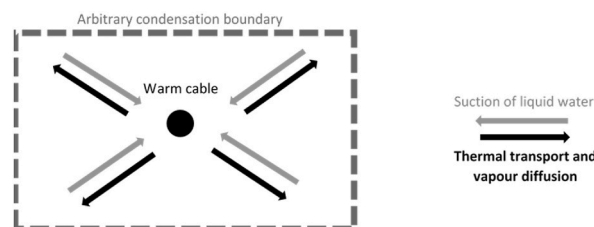


Fig. 1. Simplified systematic sketch for coupled heat and moisture transport in the soil surrounding an operating cable.

on the vapour diffusion. A high  $\lambda_{app}$  gives lower temperature, and thermal gradient under steady-state conditions and less tendency for the soil to dry out. Grain size distribution of the cable sand highly affects the water retention capacity, especially the amount of fine-grained material, which controls the water content in the soil in general and close to the cable in particular. A soil with a high water retention capacity is less prone to dry out.

Experimental and numerical studies have earlier been performed to quantify heat dissipation from underground cables, and in some cases also to predict when the close surroundings of the cable may dry out [9–14]. Different types of backfill material are well studied, also the beneficial effects from using artificial backfill materials. In some cases, the moisture content was also controlled by regulating the matric suction in the sample [9,10,15,16]. That approach helps to systematically study the role of moisture content between samples with equal density. Previous studies are thorough and show the development of a dry out with constant loading, as well as under cyclic loading [13,17,18]. However, the experimental setups did not appear to allow for suction of liquid water for balancing moisture migration. Thereby, a dry out is the predetermined outcome.

The present paper presents a laboratory experiment in which the physical processes around an operating cable are investigated. In contrast to prior work, the proposed experimental setup allows for the liquid water suction to take place by capillarity, and thus mimic real conditions in the field. In addition, suction is also locally measured rather than controlled. Furthermore, a finite element model is developed for detailed interpretation of the experiment, such as the effect of high temperature and low water content on the apparent thermal conductivity.

## 2. Experimental methodology

### 2.1. Test setup

In the present experimental setup, a steel sample holder with a height of 180 mm and diameter of 320 mm was equipped with an electrical heating rod  $\text{Ø}8 \times 320$  mm (with a built-in thermocouple) in the center of the sample holder, a ceramic plate with a thickness of 10 mm, a tensiometer  $\text{Ø}23$  mm and type-K thermocouple sensors, range  $0^\circ\text{C}$  to  $+400^\circ\text{C}$ , accuracy  $\pm 0.4^\circ\text{C}$ , diameter 0.2 mm. The sample holder was insulated by a 70 mm thick insulation material of mineral wool, except for the bottom of the sample holder, where 20 mm insulation of cellular plastic was placed. The sample holder was connected to a vacuum pump, for control of the groundwater level, see Fig. 2.

Two cable sand materials were used in this experiment, Södra Sandby (SS) with grain size 0–5 mm,  $d_{50}$  0.5 mm and dry density  $1976 \text{ kg/m}^3$  and Hamneda (H) with grain size 0–4 mm,  $d_{50}$  0.45 mm, and dry density close to  $1800 \text{ kg/m}^3$ . The cable sand was placed in the sample holder, connected to vacuum containers. The heater mimicked a cable in operation. The groundwater level was controlled by suction in the ceramic plate. The suction in the cable sand was measured by a tensiometer 5 mm from the heating rod, see Fig. 2. The temperature close to the heating rod was measured in three points by thermocouples type K, spaced horizontally 5, 10, and 20 mm from the center of the horizontal heating rod, but with 10 mm offset from each other and vertically in line with the heater. Furthermore, temperatures for the steel side, steel lid of the sample holder, and ambient temperature were also measured and sampled with the logger, dataTaker DT85.

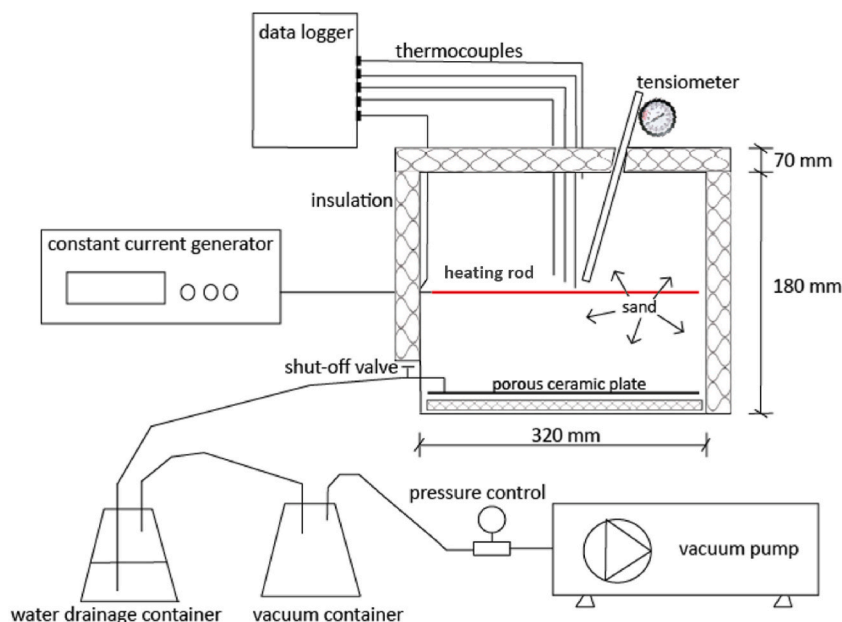


Fig. 2. Experiment setup.

## 2.2. Measurement procedure and results

The sand material in the sample holder was compacted according to standard proctor compaction. Thereafter, the samples were saturated with tap water from the bottom and up for one night (12 h). The required power in the heating rod was unknown. Thus, the temperature of the heating rod was stepwise raised by electrical power with a constant resistor, up to the desired temperature level of approximately 75 °C. In the next step, the “ground” water level was decreased by using the vacuum pump, and stepwise increasing the suction to −5 kPa, −10 kPa, −20 kPa, and −40 kPa, which correspond to groundwater levels of −0,5 m, -1m, -2m and -4m below the heating rod. Groundwater levels vary with climate conditions and topological conditions, mainly in recharge areas. However, there is a low variation of the groundwater level in discharge areas, with levels close to the ground surface [19]. The groundwater levels in this experiment were selected based on typical groundwater levels in Swedish moraines and sandy soils, which are 1–4 m below soil surface [20]. Finally, the water in the cable sand was first fully drained and then oven-dried at 105 °C, to establish the total dry condition.

Density, gravimetric and volumetric water content were measured after compaction, saturation, and at the end of the experiment, see Table 1. The sands fall within the range of typical coarse-grained backfill materials. The  $\lambda$  of the material was measured at three occasions by using the thermal needle probe method, which follows ASTM standard D5334-14 [21]: first after compaction (ac) of the cable sand material in the sample holders, then after water saturation (as) and finally at end of the experiment (ee). The results are presented in Table 1.

The most significant difference between the cable sands was the mineral content. For cable sand SS, there was a much larger content of the highly conductive mineral quartz compared to cable sand H, which resulted in a higher  $\lambda$ . The larger proportion of smaller grain size in cable sand H, allowed for a higher capillarity and water retention capacity compared to cable sand SS. Especially at intermediate volumetric water content. The negative matric potential on the y-axis in Fig. 3, symbolizes the groundwater level (m) below a cable. The lowest points correspond to  $\lambda$  at the saturated state, and the highest points correspond to  $\lambda$  at the end of the experiment.

## 3. Numerical modelling

A numerical model was developed in COMSOL Multiphysics for the evaluation of heat and moisture transfer in the experimental setup. The partial differential equation for heat transfer was solved in 3D, by using Eq. (1). Convective heat transfer boundary conditions have been used for all external surfaces of the sample holder, calculated by using Eq. (2). The heat transfer coefficient was calculated to 5 W/(m<sup>2</sup>•K). In the numerical model, the thermal conductivity of the cable sand is a function of temperature. Furthermore, the influence of water content and vapour diffusion on the temperature field, were treated by using a calculated  $\lambda_{app}$ , see chapter 3.1.

$$\rho C_p \frac{\partial T}{\partial t} = \lambda_{app} \nabla^2 T + \dot{q} \quad (1)$$

$$Q_c = h_c A (T_s - T_a) \quad (2)$$

In Eq. (1)  $\rho$  is density (kg/m<sup>3</sup>),  $C_p$  is specific heat capacity (J/Kg/K),  $T = T(x, y, z, t)$  is temperature as a function of space and time,  $\lambda_{app}$  is apparent thermal conductivity (W/(mK)) and  $\dot{q}$  is a heat source (W/m<sup>3</sup>). In Eq. (2)  $Q_c$  is heat loss from sample holder (W),  $h_c$  is heat transfer coefficient at external surfaces outside the insulation,  $h_c = 2 |T_s - T_a|^{1/4}$  (W/(m<sup>2</sup>K)),  $A$  is surface area (m<sup>2</sup>),  $T_s$  is the surface temperature (K) and  $T_a$  is air temperature (K).

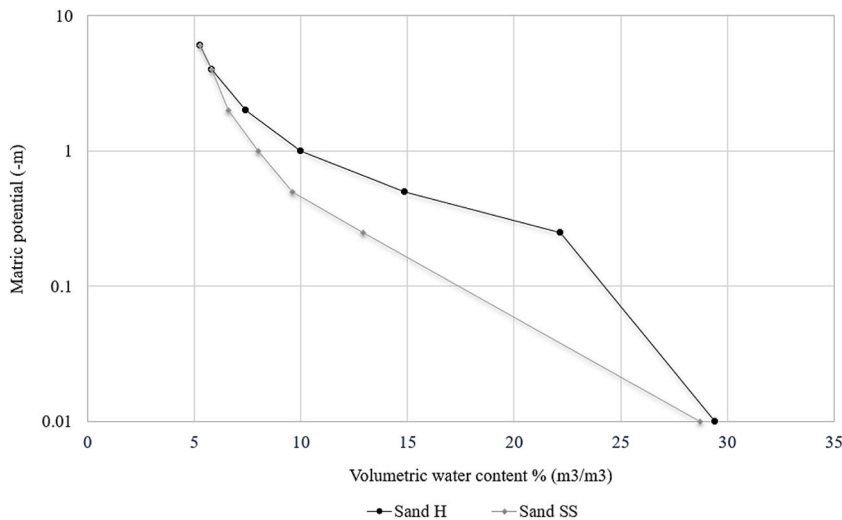


Fig. 3. Water retention curves.

### 3.1. Calculation of apparent thermal conductivity in the model

The heat transfer in partially saturated cable sand at high temperatures is very complex to model. The uncertainties are high, which some of the previous models show [22–26]. The influence of temperature and water content on  $\lambda_{app}$  were calculated using the empirical relations Eqs. (3) and (4), which were proposed by Tarnawski, Leong [27].  $\lambda_{dry}$  and  $\lambda_{sat}$  are known, see Table 1, where  $\lambda_{dry}$  corresponds to  $\lambda_{ee}$ . The temperature-dependent equations, which were chosen for their simplicity, are based on Eq. (3) by Johansen [28], which includes the Kersten function  $Ke$  [29]

$$\lambda_{app} = \lambda_{dry} + (\lambda_{sat} - \lambda_{dry})Ke \quad (3)$$

$$Ke = \frac{a + bT + cS_r + dS_r^2}{1 + eT + fS_r + gS_r^2} \quad (4)$$

where  $Ke$  is the modified Kersten number, and  $T$  is the temperature. The fitting coefficients shown in Table 2 derive from soils categorized ‘coarse’ in Tarnawski et al. (2000). This is a soil type similar to the cable sands in this lab experiment.

The calculated impact from temperature and volumetric water content, on the apparent thermal conductivity in cable sands SS and H, is presented in Fig. 4 and Fig. 5 respectively.

Figs. 4 and 5 show the variation of  $\lambda_{app}$  of cable sand as a function of moisture content and temperature, by using Eqs. (3) and (4). At the total dry state, vapour diffusion is zero and does not affect  $\lambda_{app}$ , likewise for the saturated state. The calculated results presented in Figs. 4 and 5 indicate that the influence of temperature variation on  $\lambda_{app}$  is significant when the volumetric water content is between 10% and 15%. For this span, for both cable sand SS and cable sand H,  $\lambda_{app}$  increases with 60–70% with a temperature increase from 30 to 90 °C.

### 3.2. Boundary condition in the model

The model was calibrated with respect to its outer boundary. Heat dissipation varied from the sample holder due to variations in the insulation, and due to the perforating tensiometer. For this reason, the model was calibrated by the temperature measurement results, at the heating rod as well as 5 mm and 20 mm from the heating rod, by increasing the thermal conductivity of the insulation.

Fig. 6 and Fig. 7 show a comparison between calculated temperatures by the calibrated model and the temperatures measured by the suggested setup for the two different sand types, SS and H. The calibration is based on the maximum deviation between measured and calculated temperatures, in the range 37°C–57 °C. For cable sand SS there is a small difference between measured and calculated temperatures for all compared points. The difference in heater temperature for measured and calculated temperature was 0.25 °C.

In the numerical calculations, cable sand H, in comparison to cable sand SS, has a larger but acceptable difference between measured and calibrated temperatures for all compared points. Less than 5%. The difference in heater temperature for measured and calculated temperature was approximately 1 °C. For cable sand H, it seems like the thermocouples have been dislocated a few millimeters away from the heater during the installation, and due to compaction of the cable sand.

## 4. Results and discussion

Results of temperature measurements obtained by the laboratory tests, which are compared to numerically calculated temperatures, are presented in Fig. 8 and Fig. 9, for cable sand SS, and cable sand H respectively. The measured temperatures are displayed for each cable sand, with the heating rod during the experiment (solid red in Figs. 8 and 9) as well as calculated temperatures for three different thermal conductivities (dashed lines in Figs. 8 and 9). The room temperature (solid grey) had an impact on the sample temperatures, therefore included in the figures.

For the calculated temperatures, two constant thermal conductivity levels and one thermal conductivity varying by temperature and moisture content were used. The constant thermal conductivities for cable sand SS,  $\lambda_{ac}$  and  $\lambda_{ee}$ , were 2.15 W/mK and 1.96 W/mK respectively, and for cable sand H,  $\lambda_{ac}$  and  $\lambda_{ee}$  were 1.64 W/mK and 1.11 W/mK respectively. The variable thermal conductivities  $\lambda_{app}$  were in the range of 1.92–3.85 W/mK for sand SS and 1.12–2.43 W/mK for sand H.

From the start of the experiment until approximately 1500 h, the difference in the calculated temperatures between constant and variable conductivities increases. This is shown by a comparison between the calculated temperature ( $\lambda_{app}$ ), and the calculated temperature by the constant conductivities, see Figs. 8 and 9. The increase is most distinguishable for cable sand H. This deviation is

**Table 1**  
Measured properties for cable sands H and SS at room temperature.

Sand	Status <sup>a)</sup>	Density kg/m <sup>3</sup>	Dry density kg/m <sup>3</sup>	GWC <sup>b)</sup> $m_w/m_{dry}$	VWV <sup>b)</sup> $V_w/V_{tot}$	$\lambda$ W/(mK)
H	ac	1896	1801	0.050	0.090	1.64
	as	2019	1801	0.118	0.292	1.79
	ee	1869	1822	0.025	0.046	1.11
SS	ac	2080	1976	0.050	0.099	2.15
	as	2187	1976	0.105	0.287	3.14
	ee	2111	2058	0.026	0.054	1.96

<sup>a</sup> ac: after compaction as: after saturation ee: end of experiment.

<sup>b</sup> GWC: Gravimetric water content VWV: Volumetric water content.

**Table 2**  
Fitting coefficients [27].

Coefficient	a	b	c	d	e	f	g	S <sup>2</sup> r
value	0.128	- 0.0012	0.556	1.167	-0.0074	-0.841	2.099	0.932

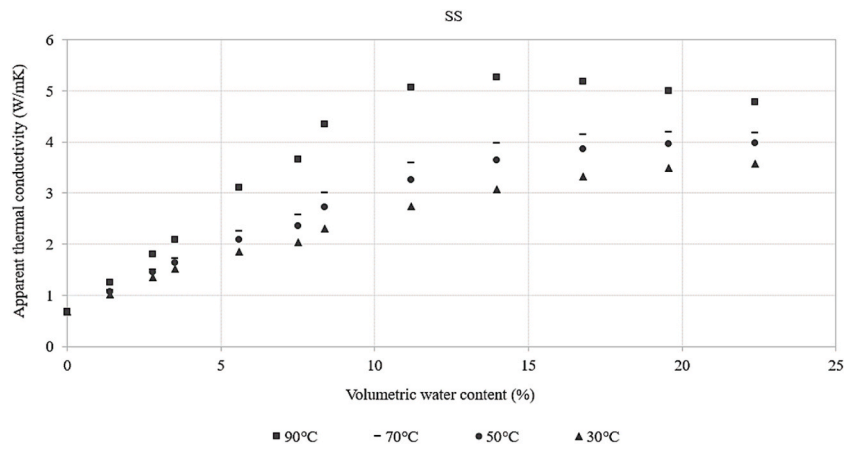


Fig. 4. Calculated  $\lambda_{app}$  by Eqs. (3) and (4) versus volumetric water content in temperature span 30–90 °C for cable sand SS.

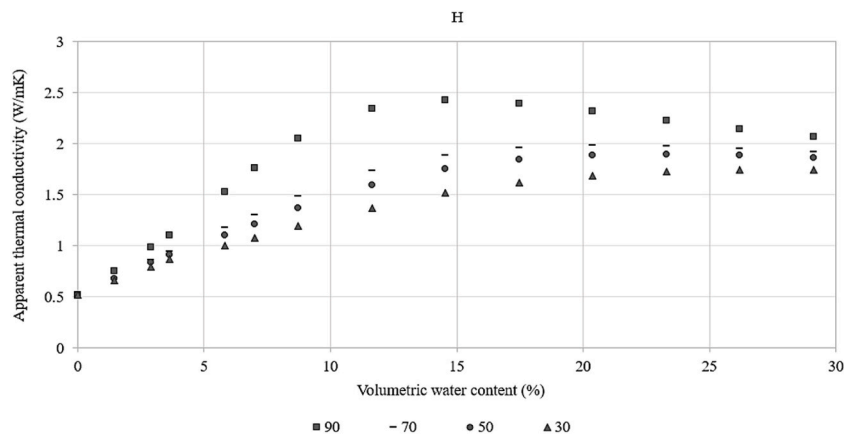


Fig. 5. Calculated  $\lambda_{app}$  by Eqs. (3) and (4) versus volumetric water content in temperature span 30–90 °C for cable sand H.

more significant after about 500 h and 55 °C. This mean that up to this level of temperature, the influence of varying thermal conductivity by temperature and moisture content is low. After about 500 h, when the difference increases furthermore, it coincides more with the experimental results from the guarded hot-plate method by Nikolaev et al. [26], i.e. a high thermal conductivity in partially saturated cable sand at high temperatures. However, the increasing level in  $\lambda_{app}$ , seems to be smaller than what was found by Nikolaev et al. [26].

The calculated temperatures with the  $\lambda_{app}$ , had a better match to heater temperature measurements of cable sand SS than to cable sand H, in the temperature range 20–55 °C. However, above 55 °C the calculated temperatures with the  $\lambda_{app}$  had a better match for cable sand H than for cable sand SS. Further research is needed to clarify the reason of this mismatch of the results, regarding certain temperature ranges and sands.

The small scale of the experiment partly affected the hypothesis of establishing a system that represents the conditions for an operative cable in a real underground installation, at shallow depth. The hypothesis relied on that the vapour condensates within the cable sand further away from the heat source, and liquid water then flows back towards the heater by capillary suction. The hypothesis of balanced vapour diffusion, and suction of water was confirmed, but in a different way than expected. Condensation did not happen due to the small scale of the experimental setup, and because water vapour diffused out through the sample holder in the opening for the tensiometer. Instead of condensation, water was transported upward from the water drainage container. Furthermore, the tensiometer had higher suction closer to the heat source in the sample holder than the created vacuum by the vacuum pump, see Fig. 10. This indicates that a water transport took place from the lower levels of the sample holder to the level of the heating rod.

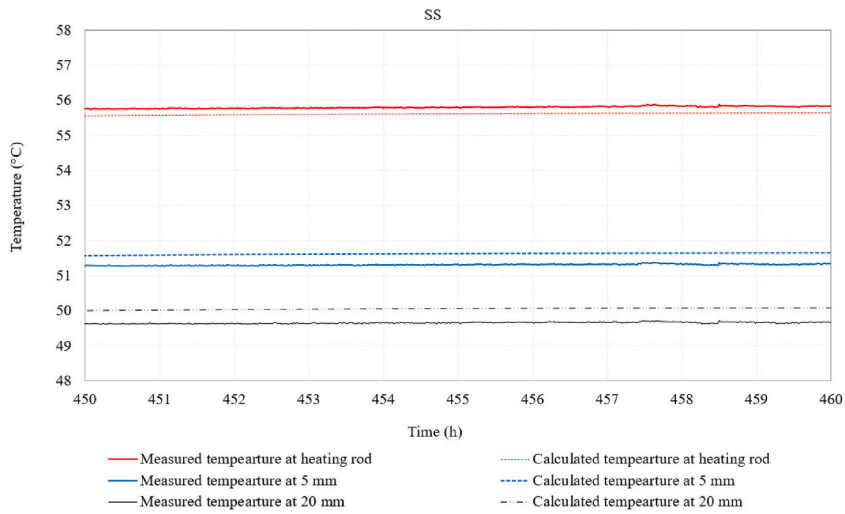


Fig. 6. Cable sand SS, comparison between measured and calculated temperatures in the temperature interval 48–58 °C.

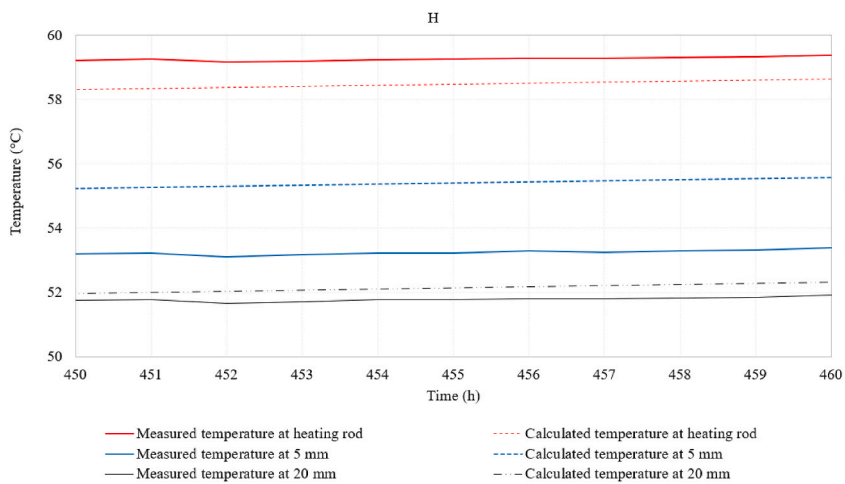


Fig. 7. Cable sand H, comparison between measured and calculated temperatures in the temperature interval 50–60 °C.

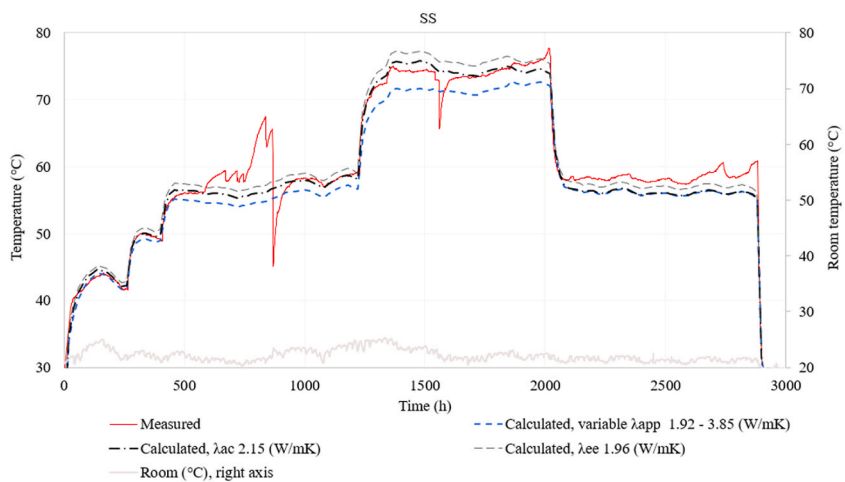


Fig. 8. Measured and calculated temperatures of heating rod for cable sand SS for 2900 h.

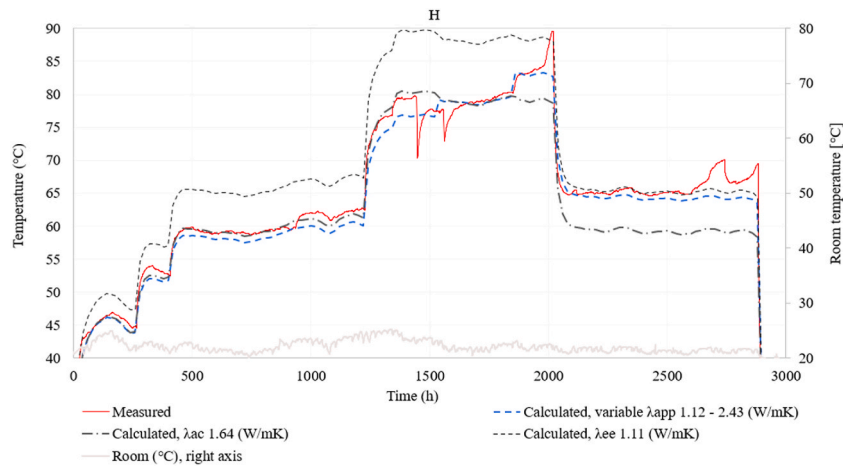


Fig. 9. Measured and calculated temperatures of heating rod for cable sand H for 2900 h.

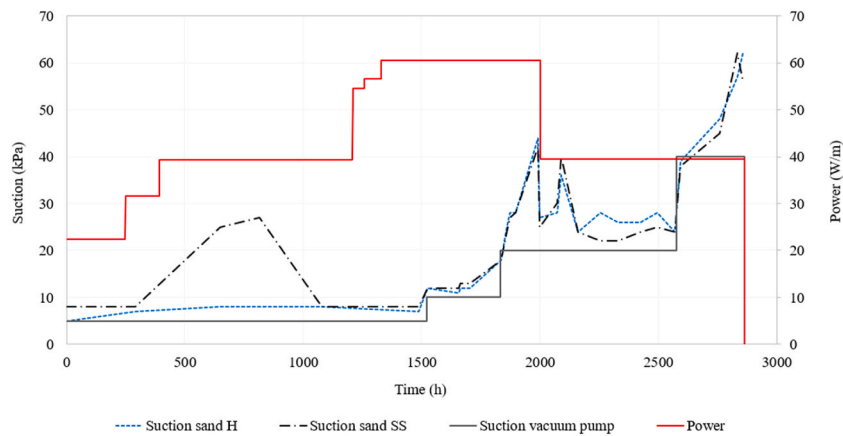


Fig. 10. Suction in cable sands H and SS from tensiometer (kPa), suction from vacuum pump (kPa), and power to the heating rod (W/m).

During the experiment, potential dry outs occurred close to the heater, only when the capillary water contact was broken. In Fig. 9 these are shown as peaks (when vacuum suction is unchanged). The broken contact, which also was visible in a transparent drainage tube, always caused fast raising temperatures in the following days. On these occasions, the heating rod was temporarily disconnected for safety reasons, which are shown by the rapid temperature drops in Figs. 8 and 9 (at 860 h and 1560 h for cable sand SS and 1425 h and 1545 h for cable sand H). The broken capillary contact may have been caused by leakage of air into the shut-off valve connections, especially when higher suctions were created with the vacuum pump. Broken water contact might also have occurred due to dehydration of the ceramic plate, caused by too high temperatures because of the insufficient distance between the heating rod and the ceramic plate.

Fig. 10 also shows the total heat flow from the heating rod in the experiment, with a range from 22.4 to 60.5 W/m, which is about 2–3 times larger than during real underground conditions of a HVDC cable. This is more than 30 times higher when measured in  $\text{W/m}^2$ , therefore a highly conservative condition. The experiment showed that even if the heating rod generated a high heat flow and high temperatures, no drying out occurred as long as there was a constant water flow towards the cable. In Fig. 10 this is shown from experimental start until 1500 h, where the suction in cable sand H was stable, even though the power reached its highest levels. First, when there was a change in vacuum suction, the cable sand responded with higher suction. The largest suction that could be created during a long period with this experimental setup, without any dry outs, was 20 kPa (groundwater level 2 m below the heating rod). This period is shown in Figs. 8–10, between 2100 h–2500 h. For this period, a suction of 25 kPa in the sands corresponds to a matric potential of  $-2.5$  m. This indicates a volumetric water content of approximately 7–8%, see the water retention curve in Fig. 3. It is unclear when the magnitude of the heat flow and the temperature, becomes more dominant than the ability of the liquid water to flow back due to the difference in suction. This cannot be illustrated in the experiment, since the observed dry outs occurred after a broken capillary connection.

## 5. Conclusions

In the vicinity of an operating cable in sandy backfill material, there are complex physical processes taking place, and backfill design plays a major role. The design can be conducted in a conservative manner, meaning calculating maximum cable conductor temperature, with surrounding backfill material in a dry state. This is a worst-case scenario, which prevents a thermal runaway and breakdown of cable insulation. However, this paper indicates that a less conservative approach may be used with knowledge of the groundwater levels, and the soil material underlying the cable bed. A less conservative approach may be taken for cable routes passing groundwater discharge areas, due to the lower variation in groundwater levels.

It can be concluded by the experimental results that:

- A high heat transport could be maintained from the cable if there was a capillary contact between the cable sand and the groundwater level.
- Avoiding high temperature close to the cable was possible, especially for cable sand SS, which contained the highly conductive mineral quartz.
- Stable temperatures were maintained for 400 h for both cable sands, even though the electrical load was 2–3 times higher than typical design values, measured in W/m, and more than 30 times higher when measured in W/m<sup>2</sup>. These temperatures remained stable even though the groundwater level was 2 m below the cable.
- The numerical model could illustrate that high temperatures and low water content contributed to a higher apparent thermal conductivity in the cable sand. Thus, it may be possible to account for increased heat transfer, which then would allow for a higher ampacity in the cable, without exceeding design temperature.

The desired condition, with no thermal runaway, is expected to be fulfilled when the following criteria are taken into consideration in the backfill design: proper compaction and high water retention capacity of the cable sand material, and high enough capillarity of the natural soil material down to the groundwater level.

For further research, the experimental setup can be improved by upscaling, to allow for a larger temperature gradient within the sample, and to make room for condensation to take place in the cable sand. Moreover, additional soil types could be evaluated.

## Author statement

We hereby declare that we are the authors of this manuscript.

## Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

## Data availability

Data will be made available on request.

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